

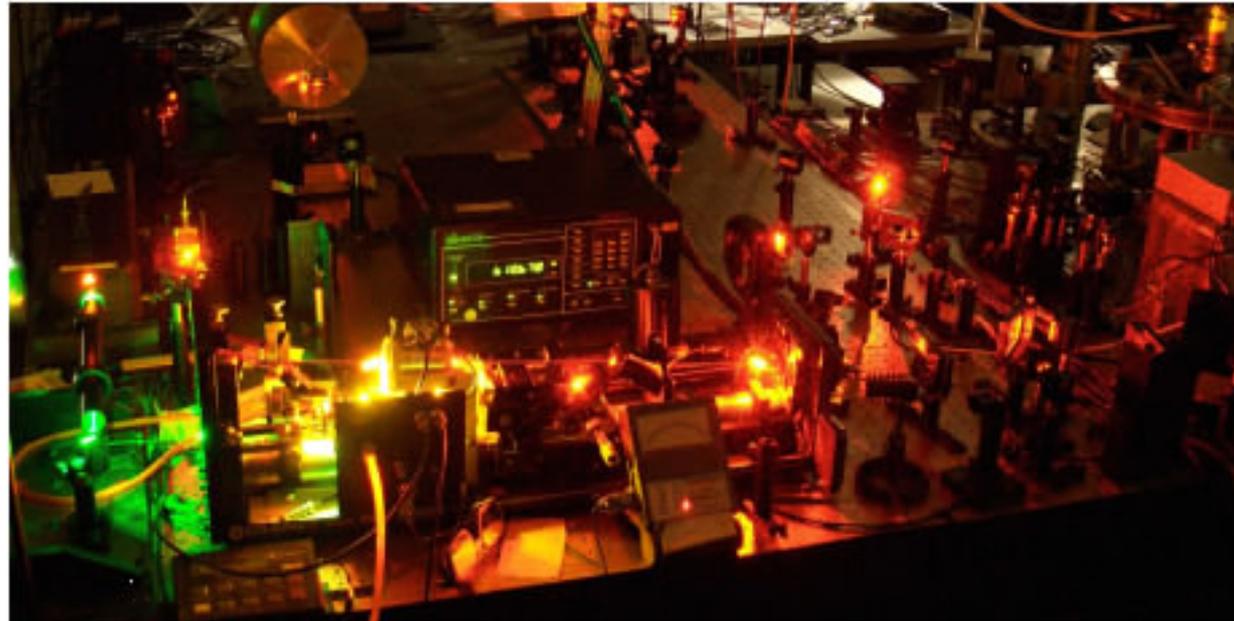
# Laserspektroskopie

**Was:** Optische Spektroskopie mit Lasern:  
Grundlagen und Anwendungen

**Wann:** Mi 13<sup>15</sup> - 14<sup>00</sup>  
Fr 10<sup>15</sup> - 12<sup>00</sup>

**Wo:** P1 - O1 - 306

**Wer:** Dieter Suter  
Raum P1-O1-216 Tel. 3512  
Dieter.Suter@uni-dortmund.de  
<http://e3.physik.uni-dortmund.de>



# Inhaltsverzeichnis

---

- 1. Einleitung**
- 2. Laser**
- 3. Zweiniveaumatome**
- 4. Experimentelle Techniken**
- 5. Laserkühlung**
- 6. Nichtklassisches Licht**
- 7. Nichtlineare Laserspektroskopie**

**W. Demtröder : Laserspektroskopie  
Springer-Verlag, 4.Auflage (2000).**

**F.K. Kneubühl, M.W. Sigrist : Laser  
Teubner Studienbücher Physik (1999).**

**D. Meschede : Optik, Licht und Laser  
Teubner Studienbücher Physik (1999).**

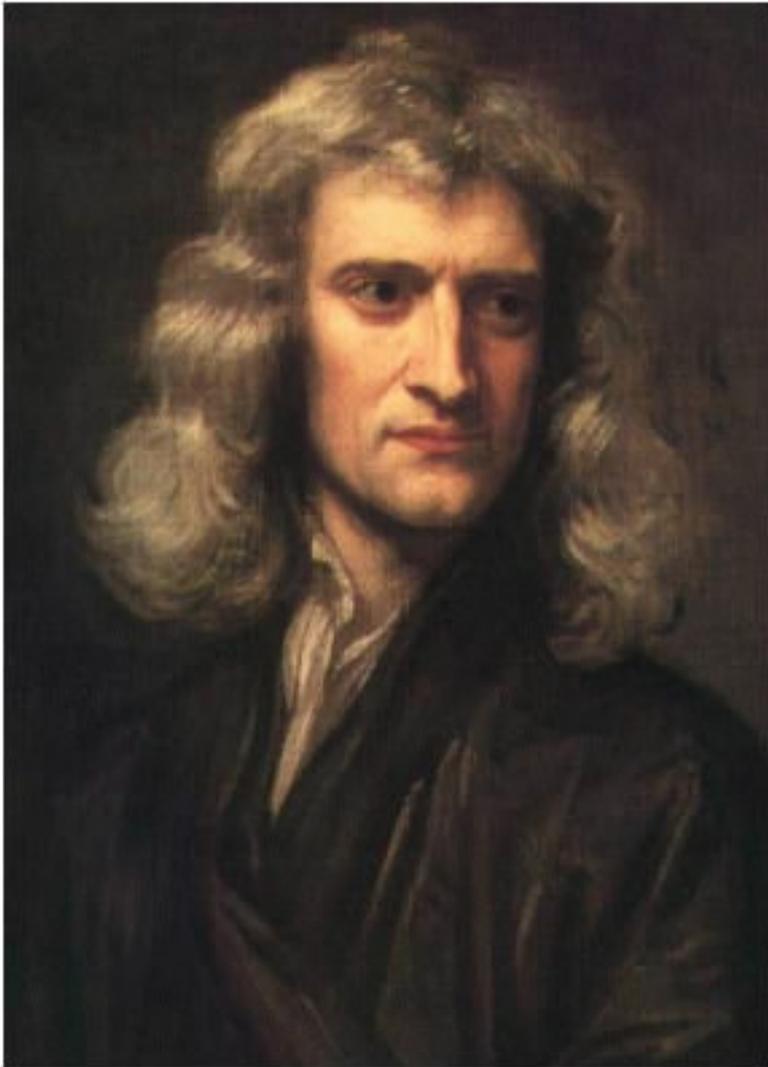
**A. E. Siegmann : Lasers  
University Science Books, 1986.**

**J.H. Eberly : Optical resonance and two-level atoms  
Dover Publications, 1975.**

**D. Suter : The physics of laser-atom interaction  
Cambridge University Press, 2005.**

# Licht und Materie

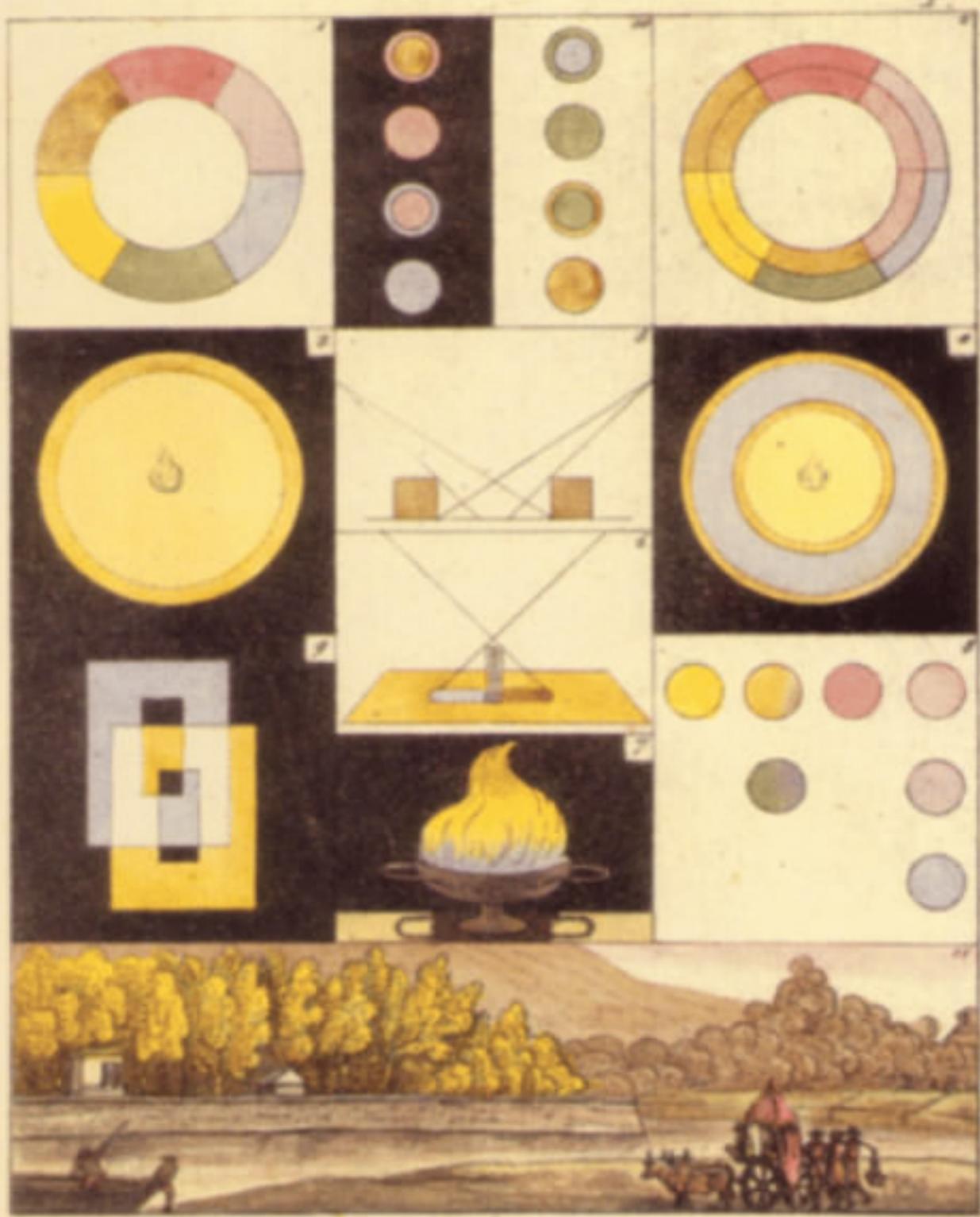
---



**Newton (1643 – 1727)**



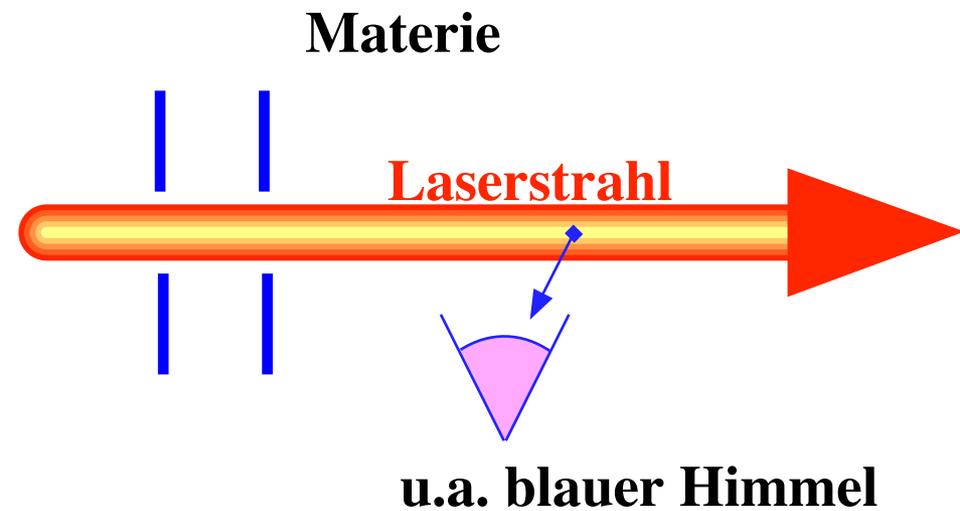
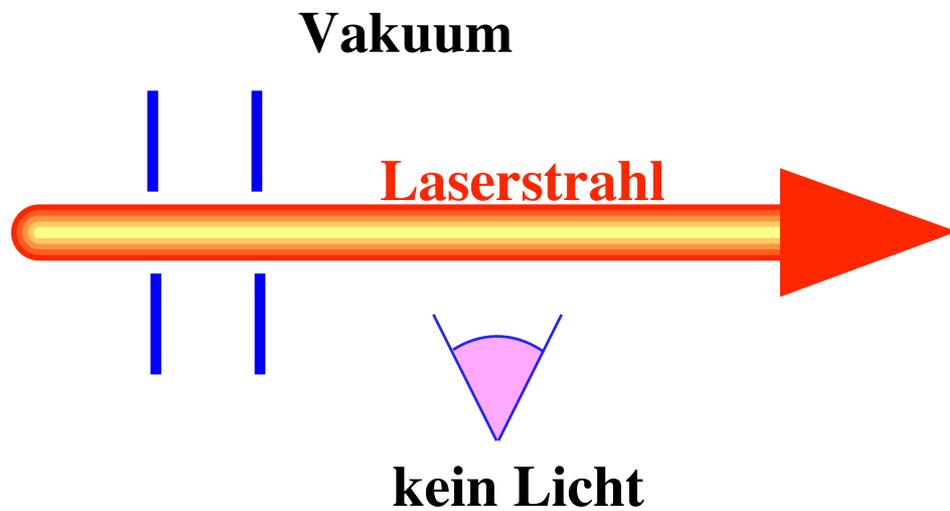
**Goethe (1749-1832)**



# Streuung



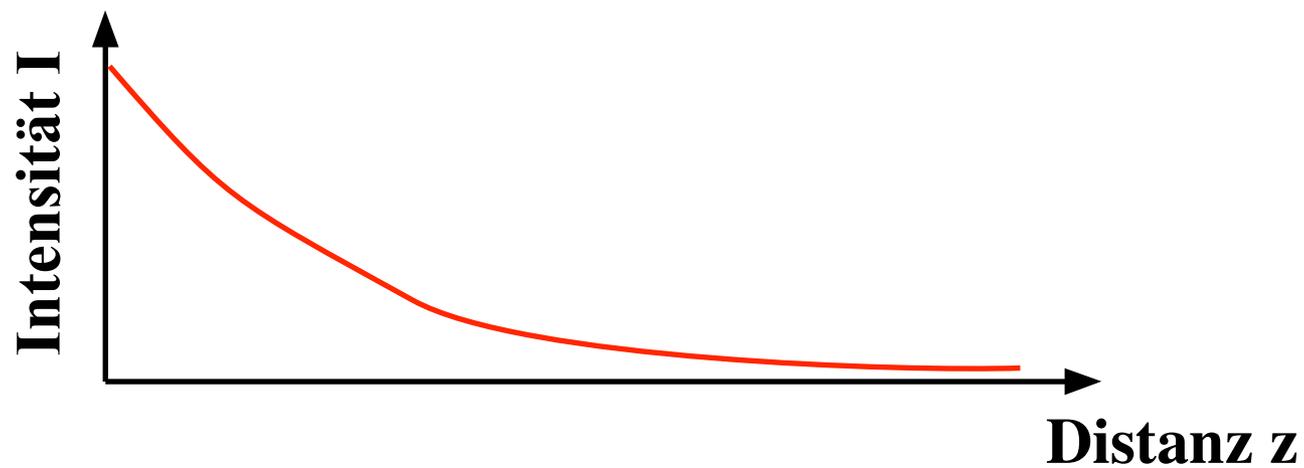
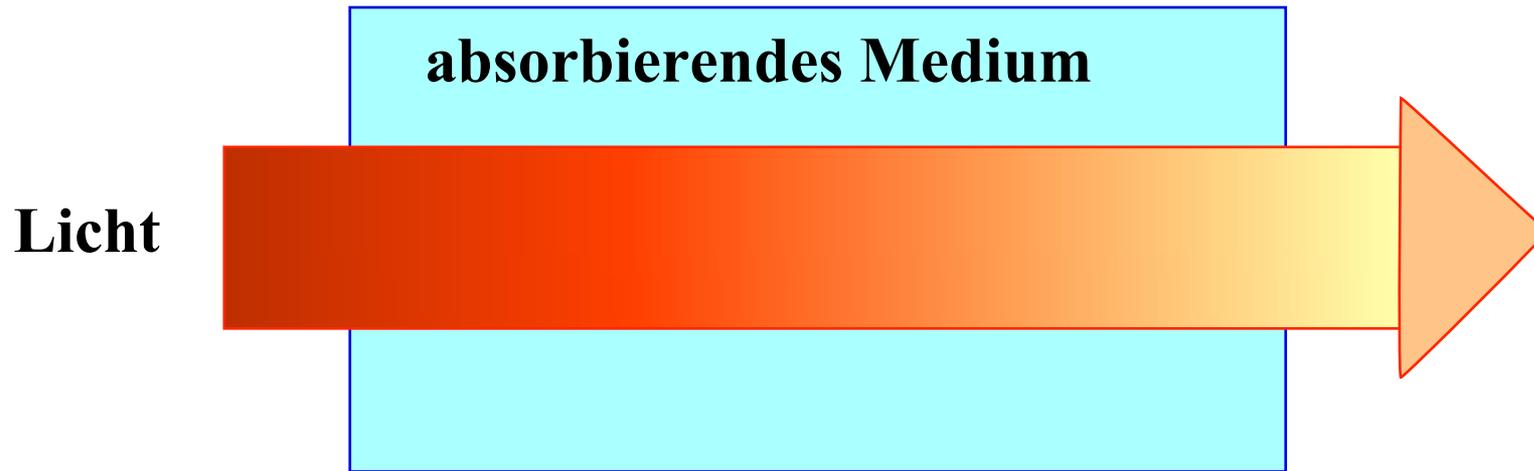
# Streuung



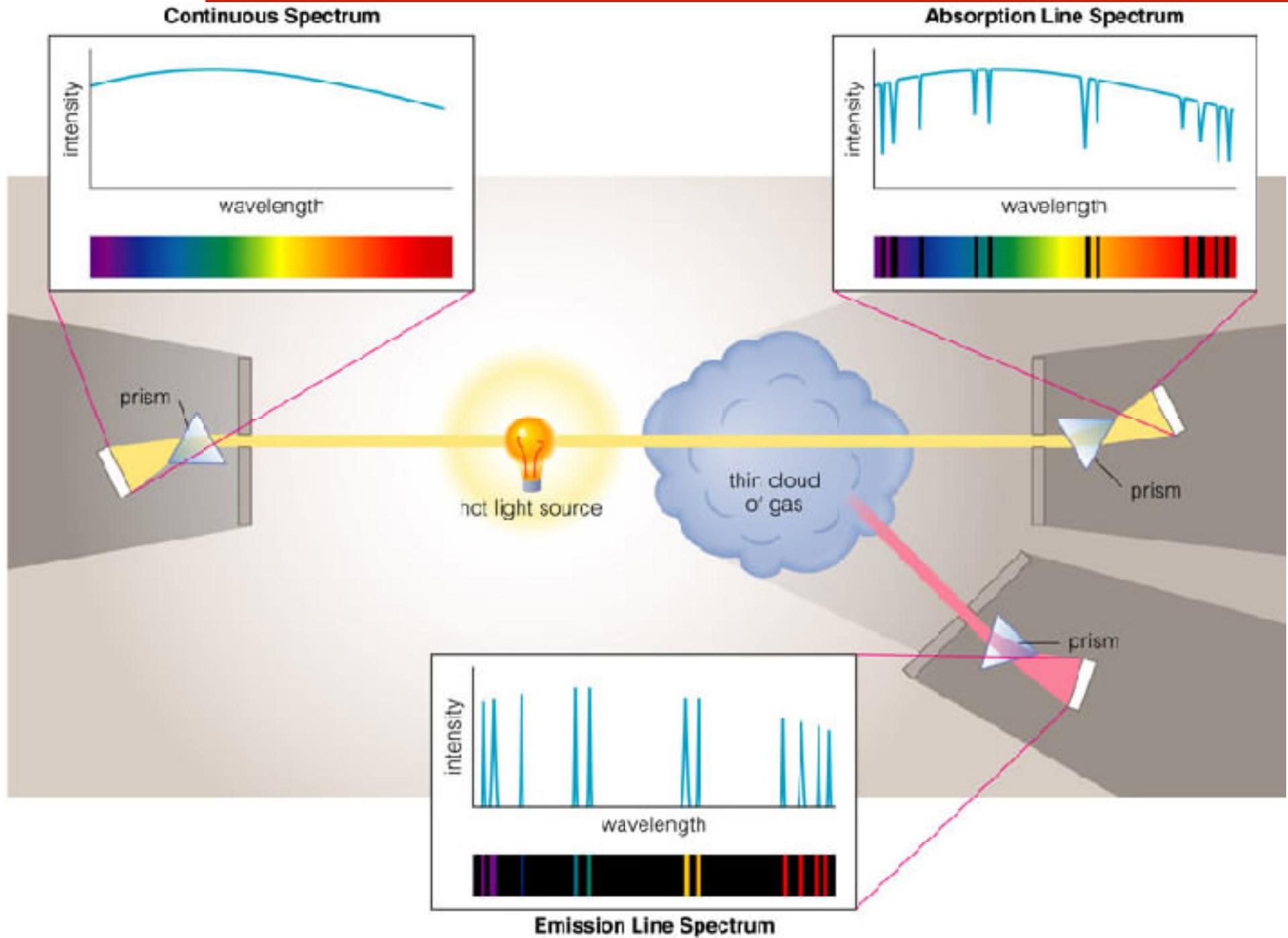
- **Homogene Materie**
- **Inhomogene Materie**

# Absorption

---

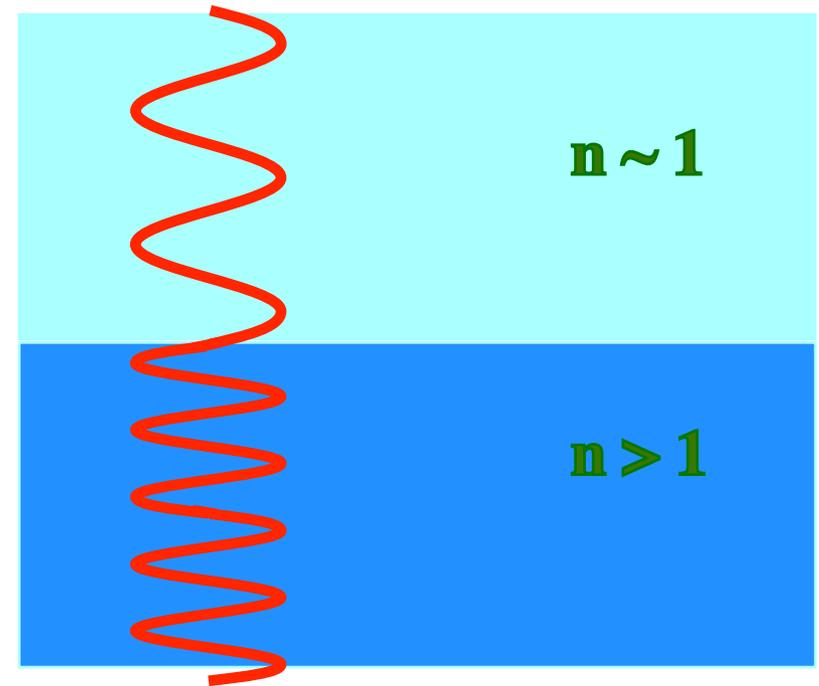
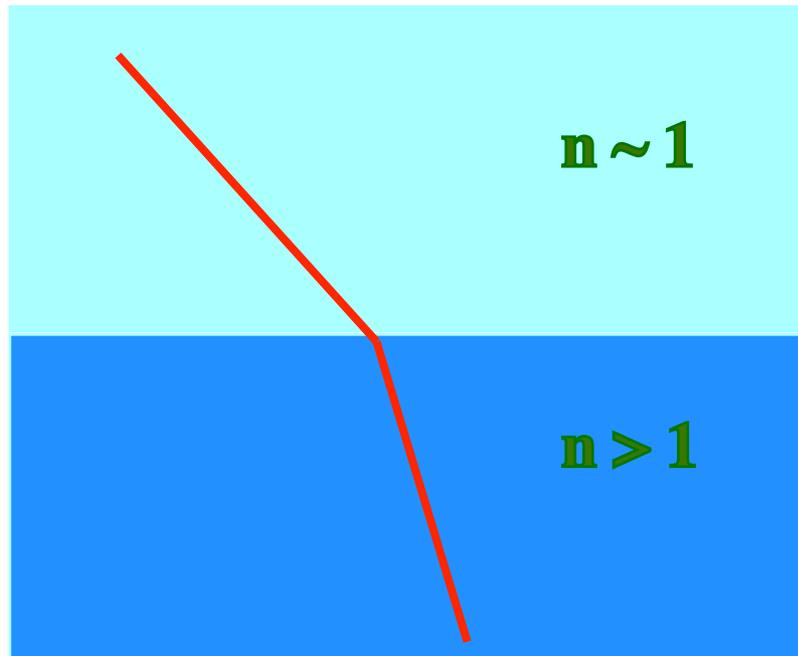


# Absorption und Emission

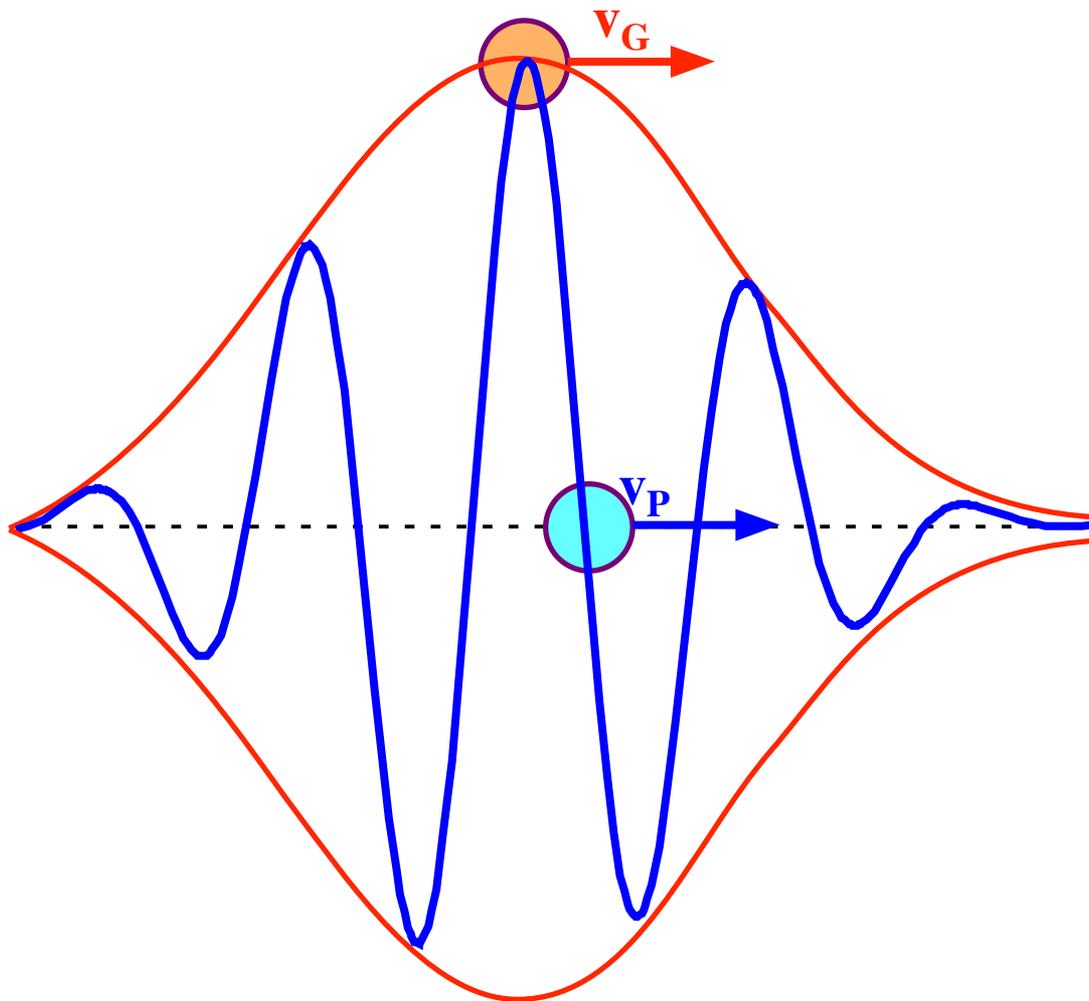


# Brechung

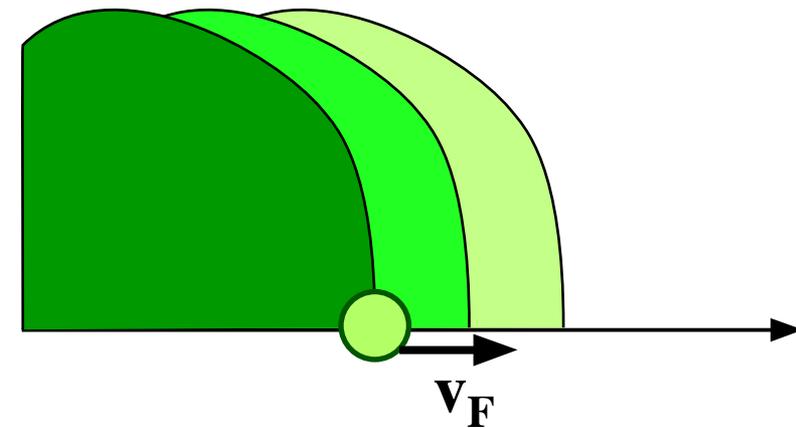
---



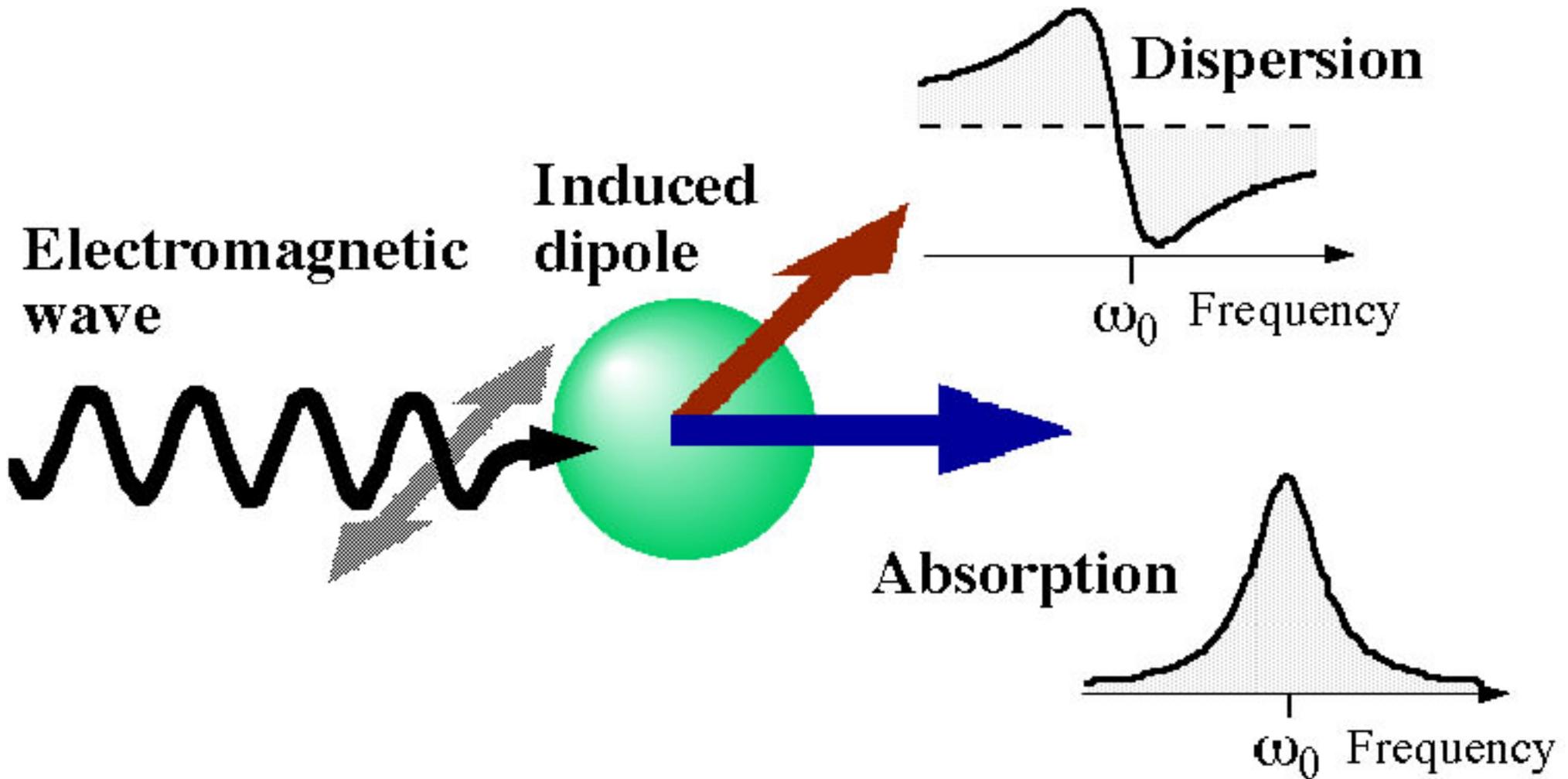
# Ausbreitungsgeschwindigkeit



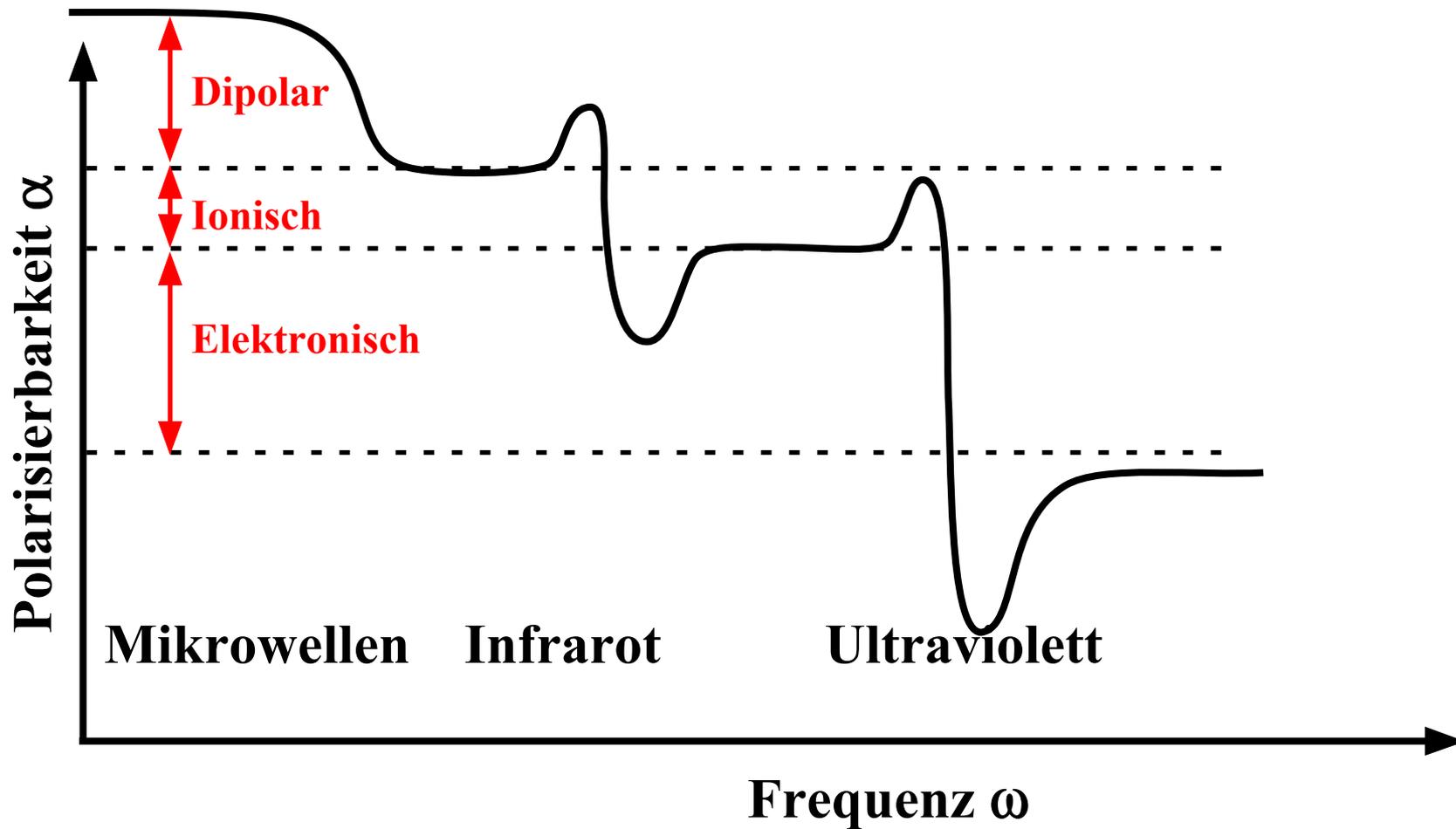
Signalausbreitung



# Mikroskopisches Modell



# Wellenlängenabhängigkeit



# 1.4 Formen der Spektroskopie

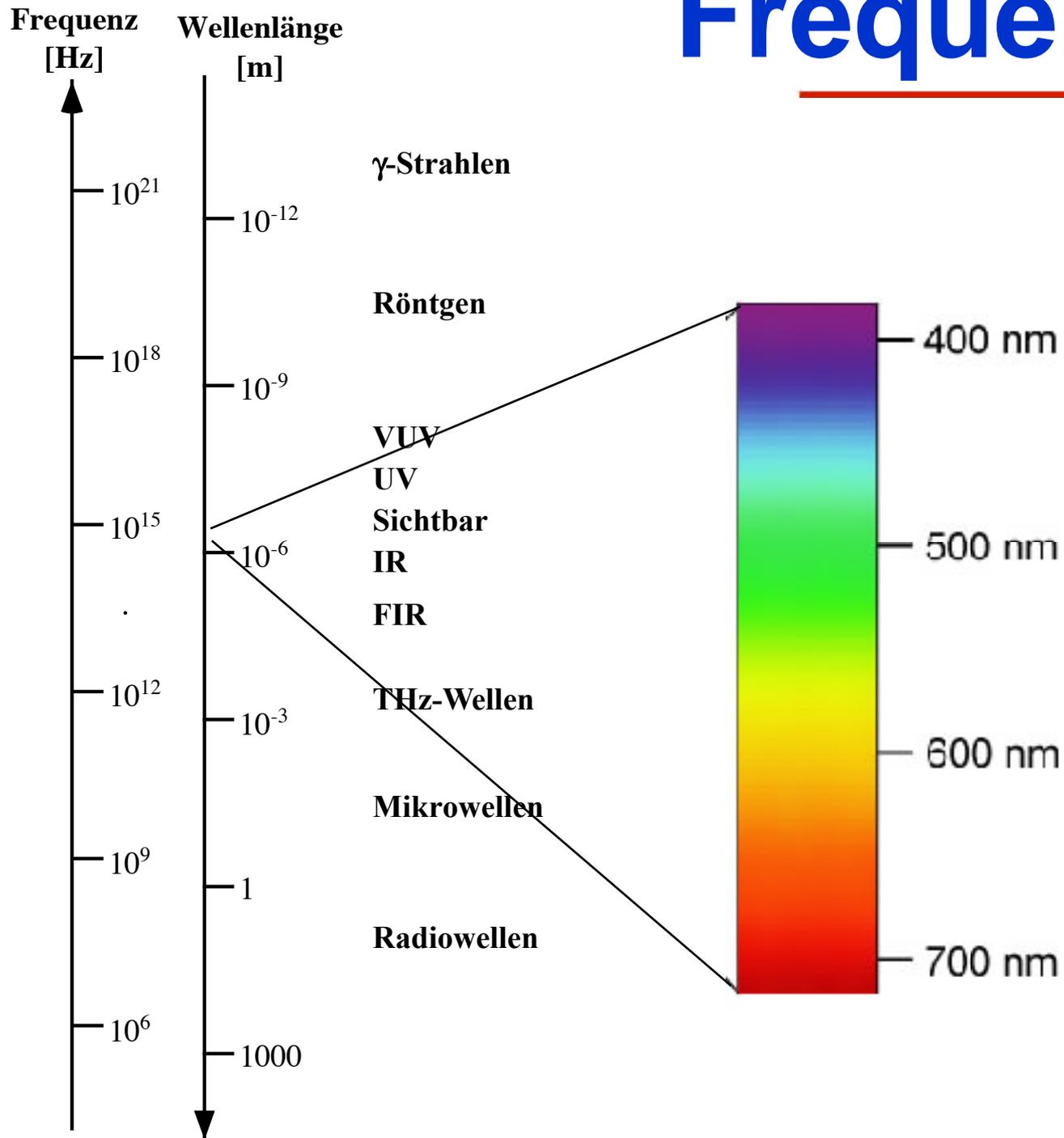
---

*Das allgemeine Umfeld:*

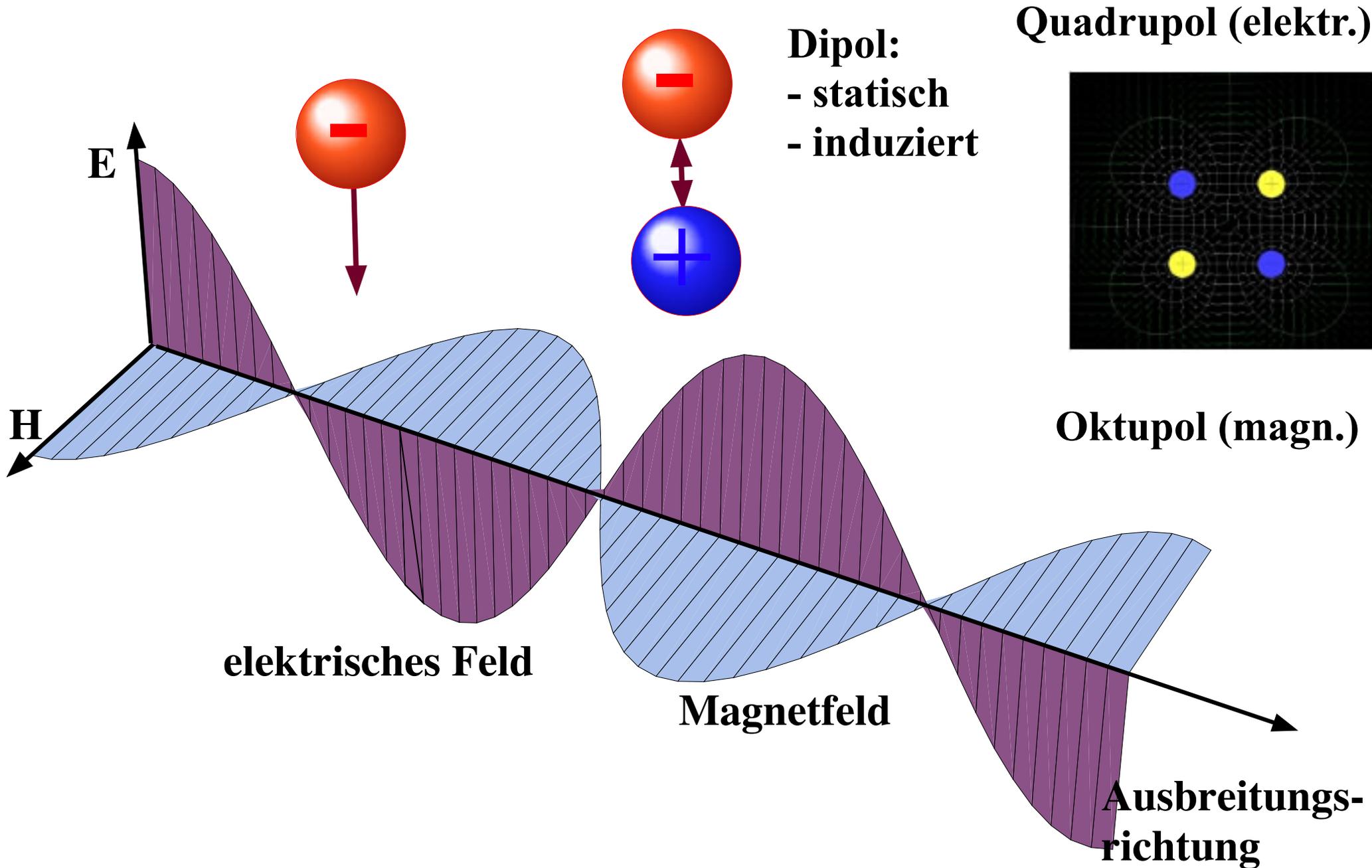
- **Wellen oder Teilchen**
- **elektrische oder magnetische Wechselwirkung**
- **Frequenzbereich**
- **lineare / nichtlineare Spektroskopie**
- **Zeit- oder Frequenzbereich**
- **Absorption / Dispersion / Emission**



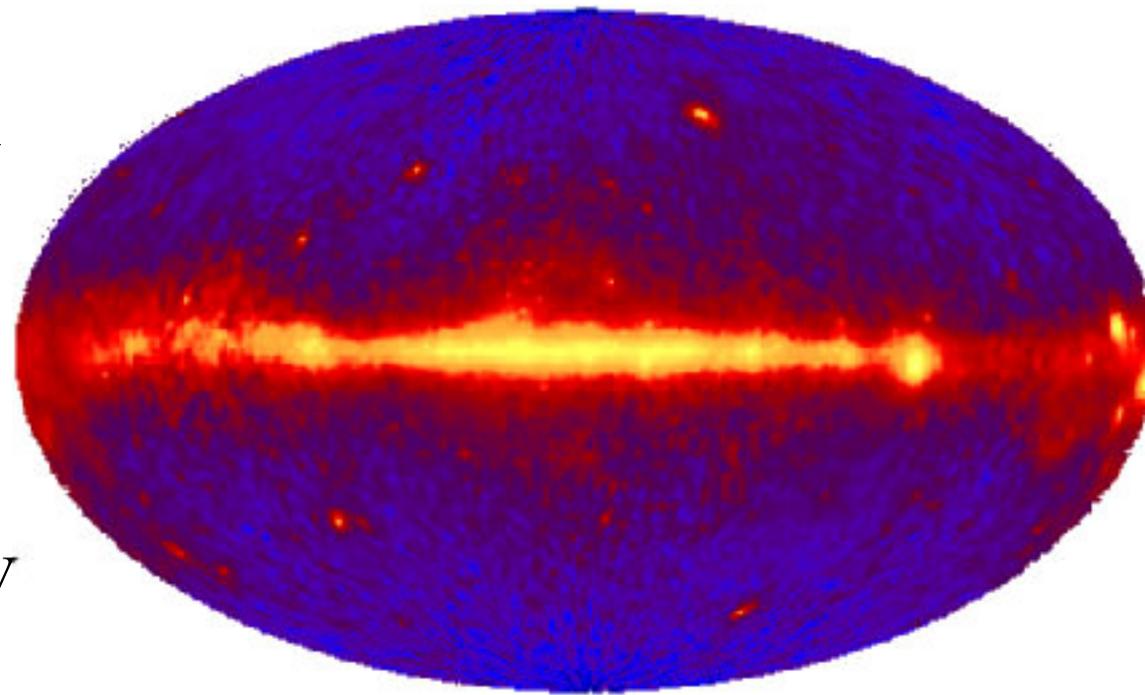
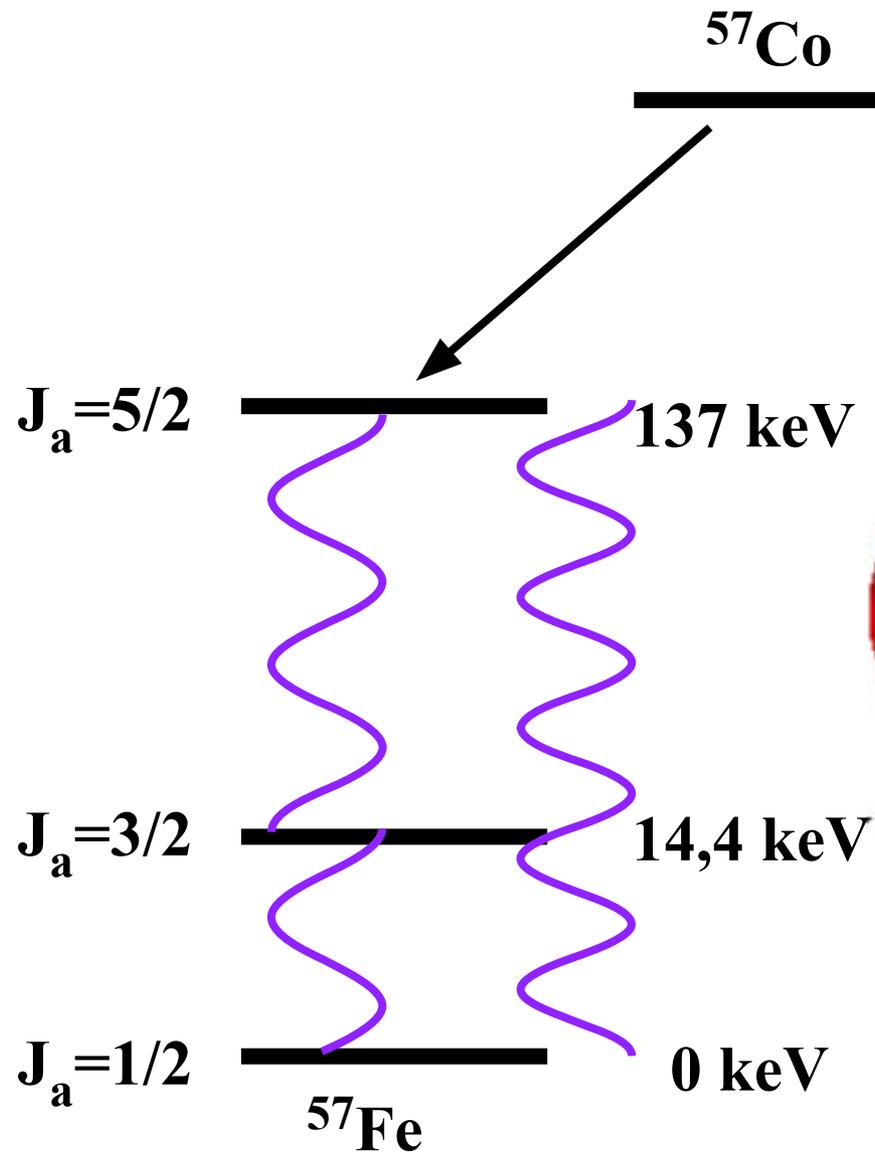
# Frequenzbereiche



# Wechselwirkungen



# $\gamma$ -Strahlen

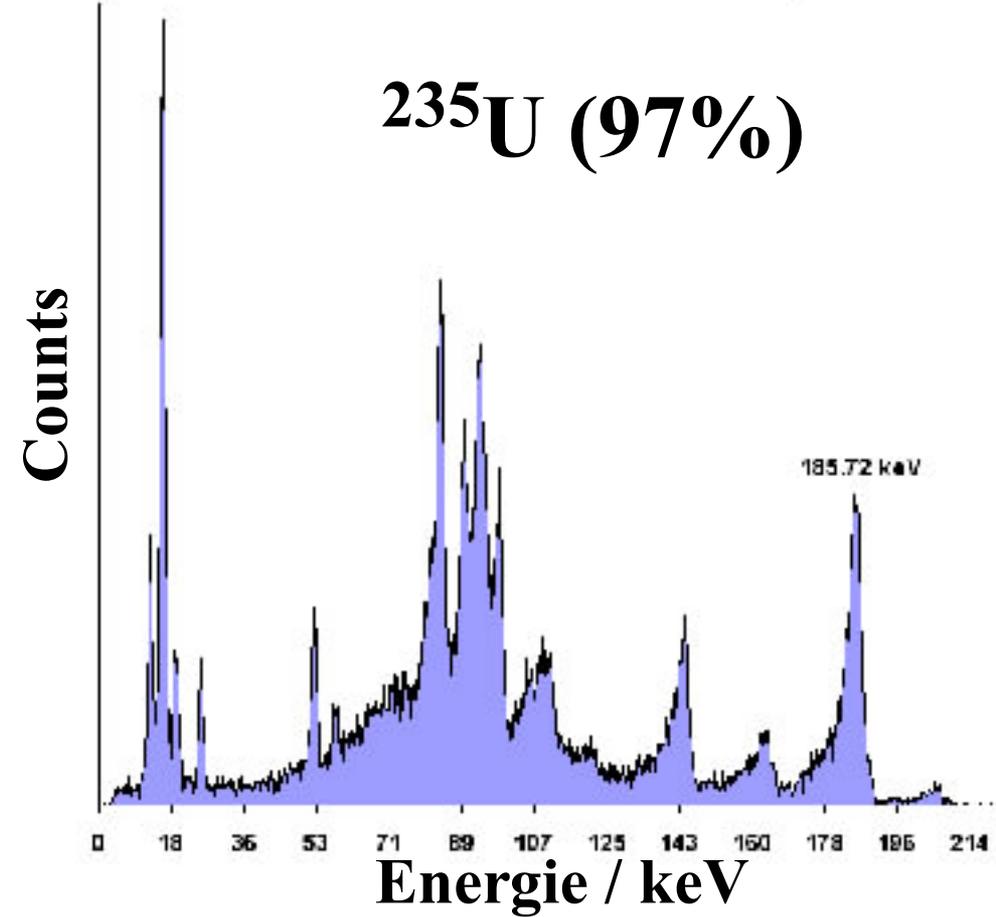
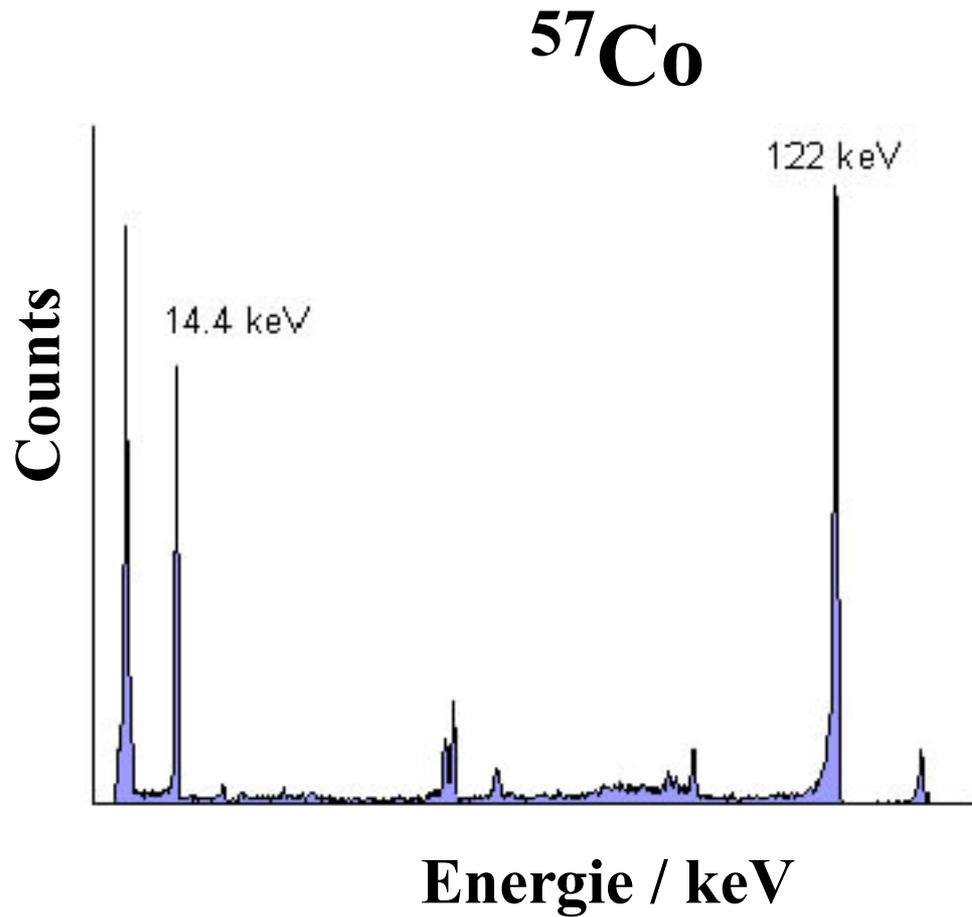


$\gamma$ -Astronomie

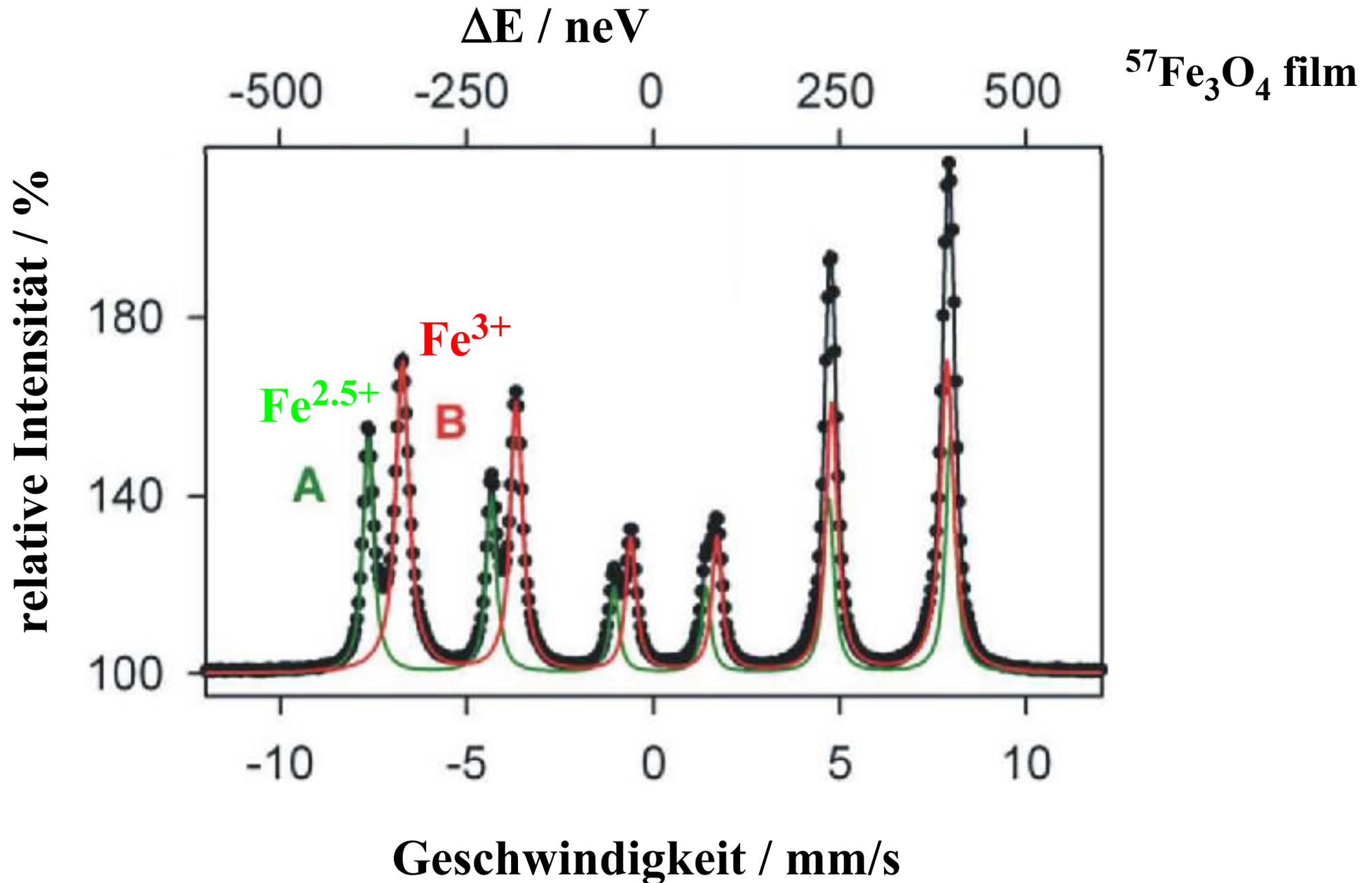
# $\gamma$ -Spektroskopie

Energie [keV]	relative Stärke	Energie [keV]	relative Stärke
$_{31}\text{Ga}^{68}$		$_{31}\text{Ga}^{74}$	
Ann. Rad.		53.92 $\pm$ 0.4	
578.3	0.04 %	380	2†
805.8	0.09 %	500	11
1077.1 $\pm$ 1	3.2 %	595.6 $\pm$ 0.2	87
1261.3	0.1 %	600	13
1764.5	0.01 %	720	2
1884.5 $\pm$ 1	0.13 %	870	5
2338	0.001 %	980	4
		1110	5
		1200	6
		1330	5
		1460	6
173	0.15 %	1560	2
1042	0.5 %	1700	3
1215		1760	3
		1930	6
		2350	15

# $\gamma$ -Spektroskopie

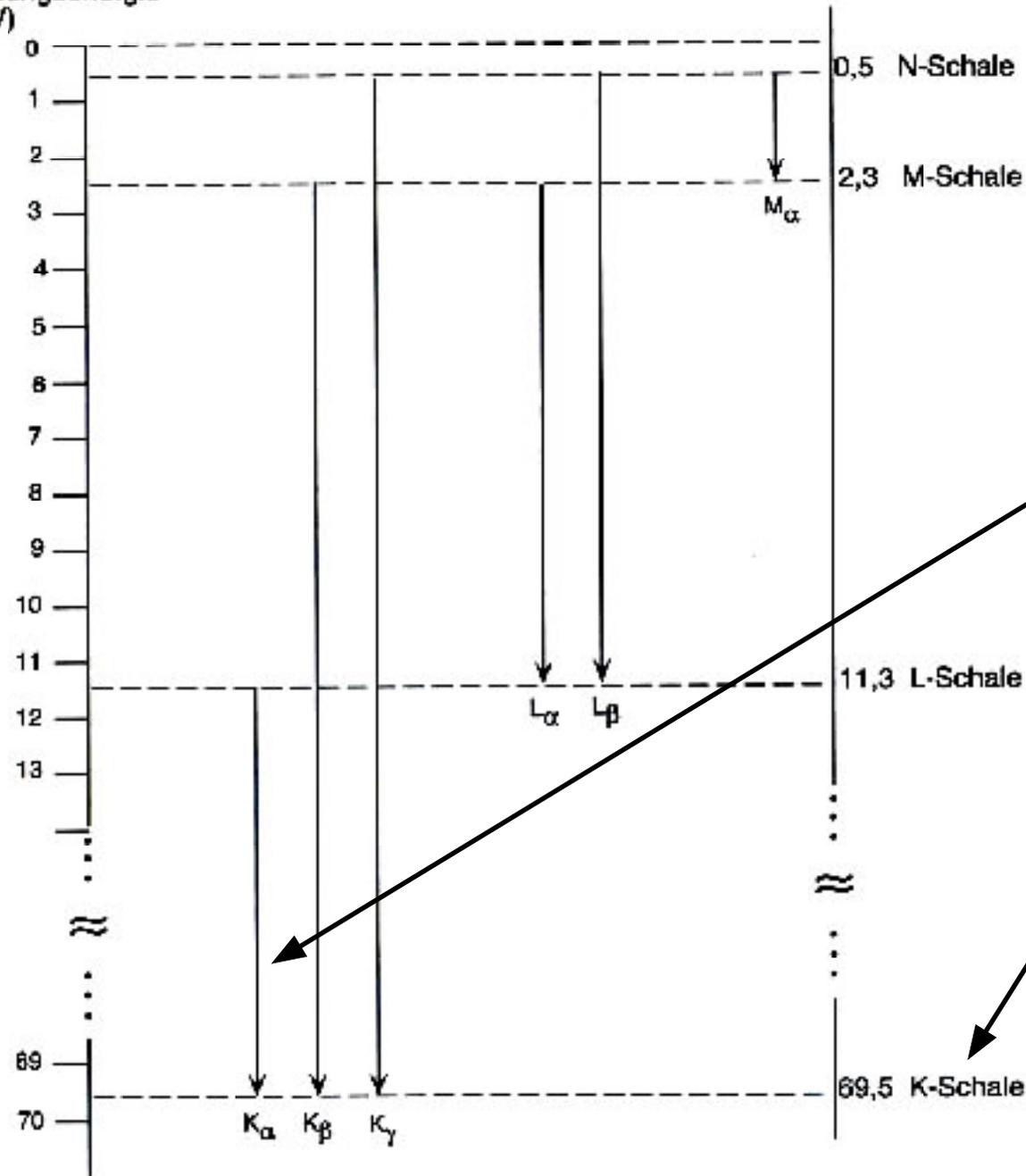


# Mössbauer-Spektroskopie



# Röntgenfluoreszenz

Bindungsenergie  
(keV)



**Bsp.: Wolfram**

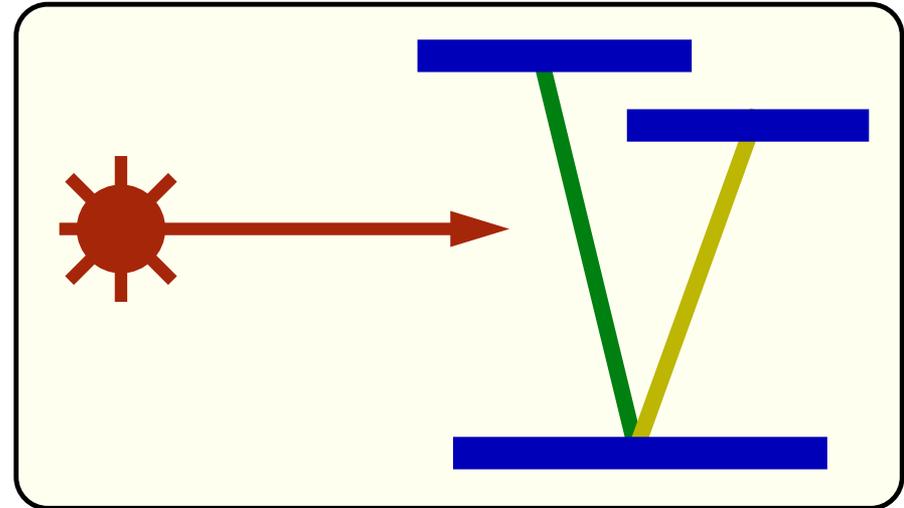
**$K_{\alpha}$ -Strahlung  $\approx 58$  keV**

**Minimale  
Anregungsenergie**

**$E \approx 70$  keV**

# Laserspektroskopie

Anregung von Valenzelektronen durch optische Photonen

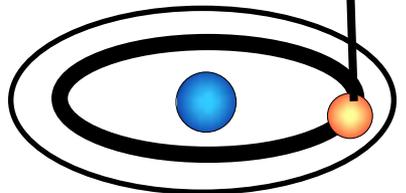
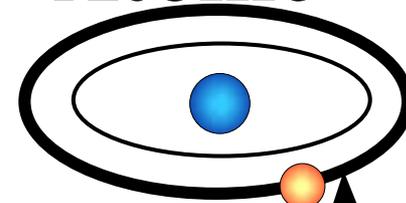


## Zustände und Systeme

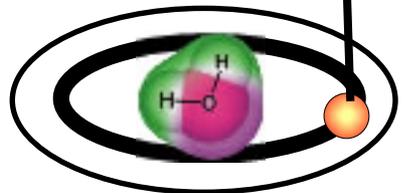
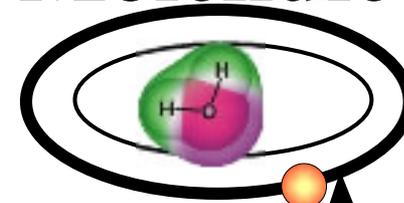
$|\epsilon\rangle$  —

$|\sigma\rangle$  —

Atome



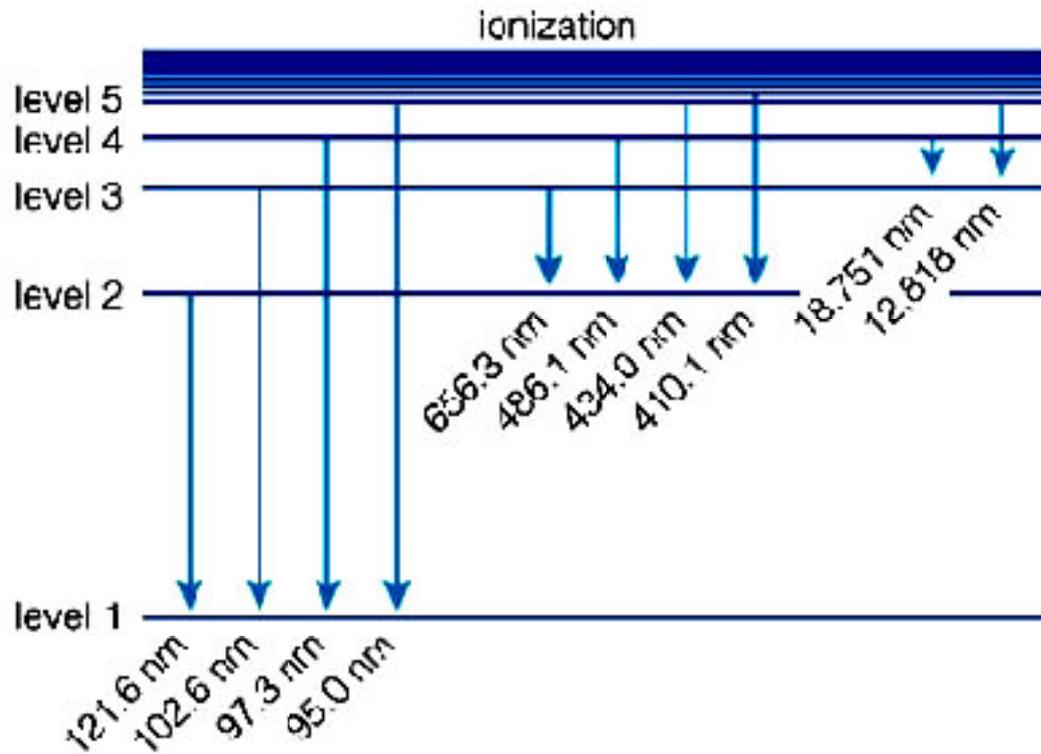
Moleküle



Halbleiter



# Beispiel : Wasserstoff



**Emission**



**Absorption**

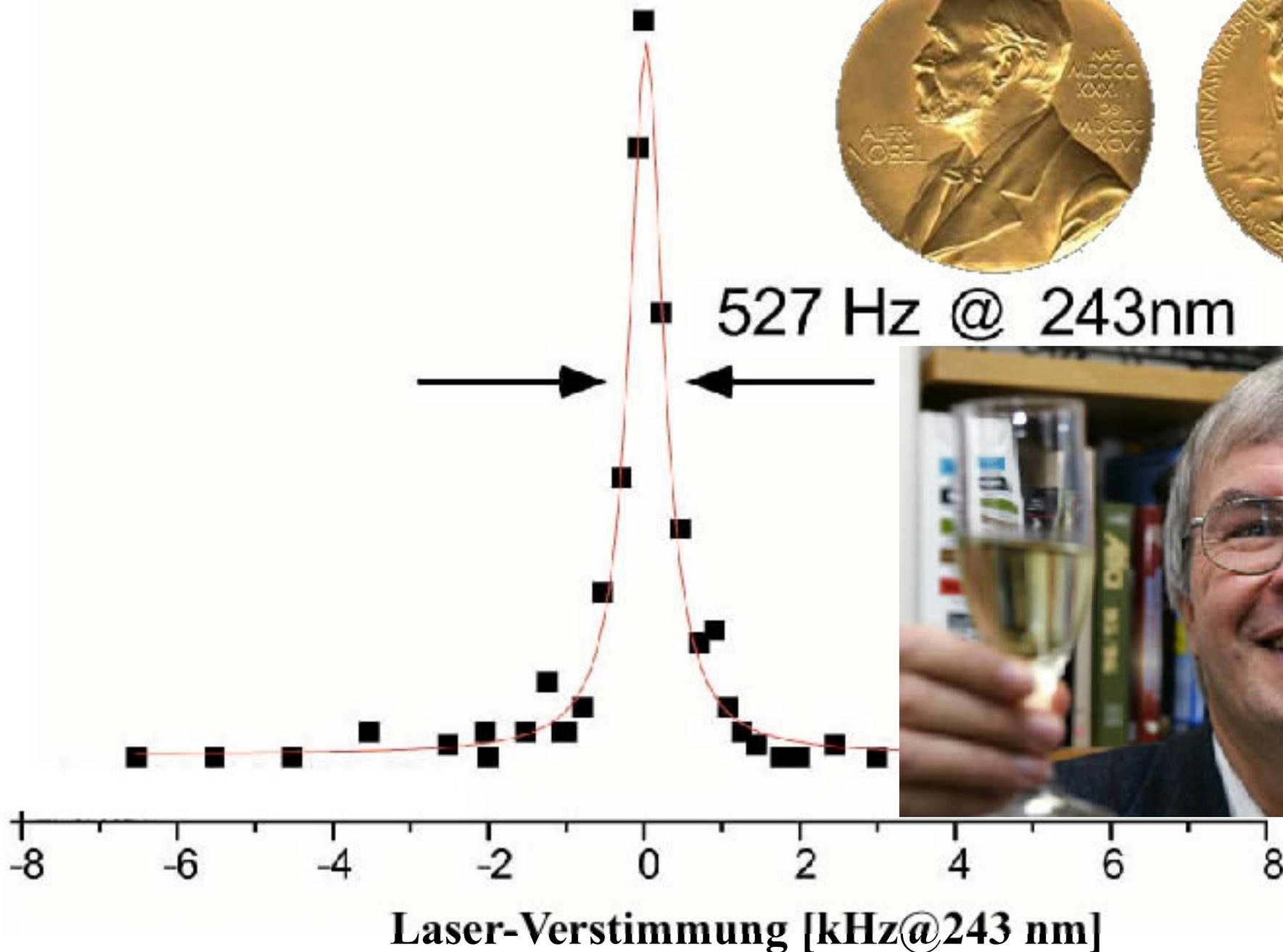


# Linienbreiten

$^1\text{H}$  1s-2s Übergang



527 Hz @ 243nm

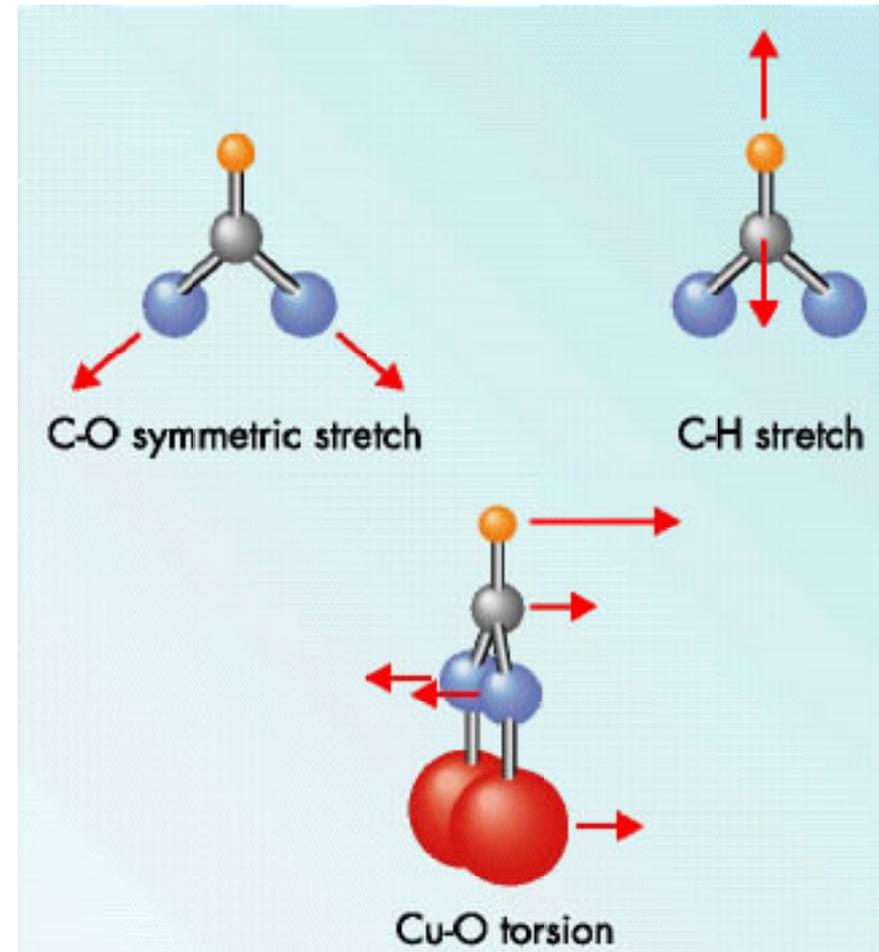
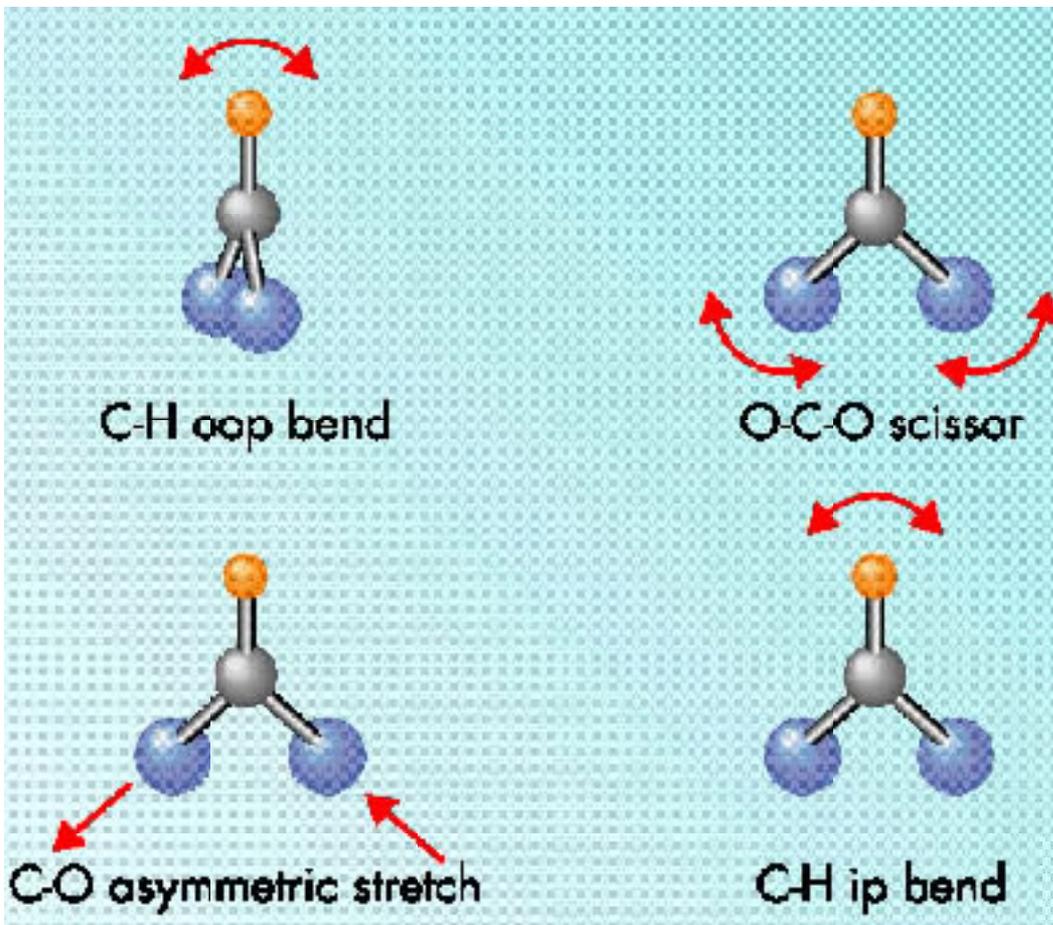


# Vibrationsübergänge

Klassischer harmonischer Oszillator:

$$\frac{m_e}{m_p} \approx 1836$$

$$\omega = \sqrt{\frac{f}{m}}$$



# IR-Banden

Frequenz /  $\text{cm}^{-1}$

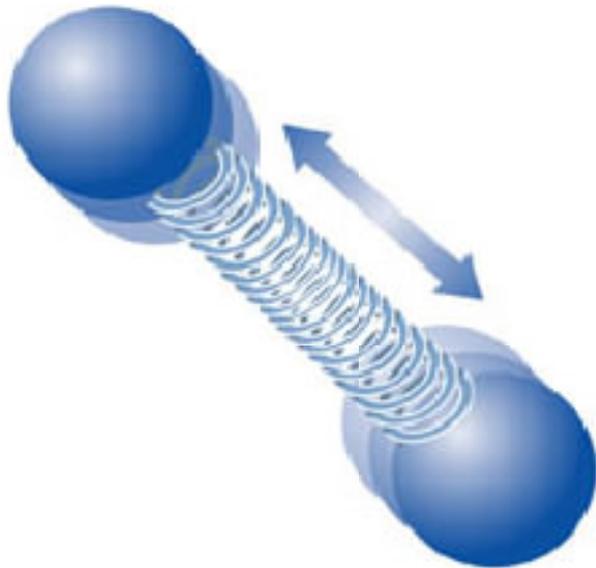
3000

2000

1000

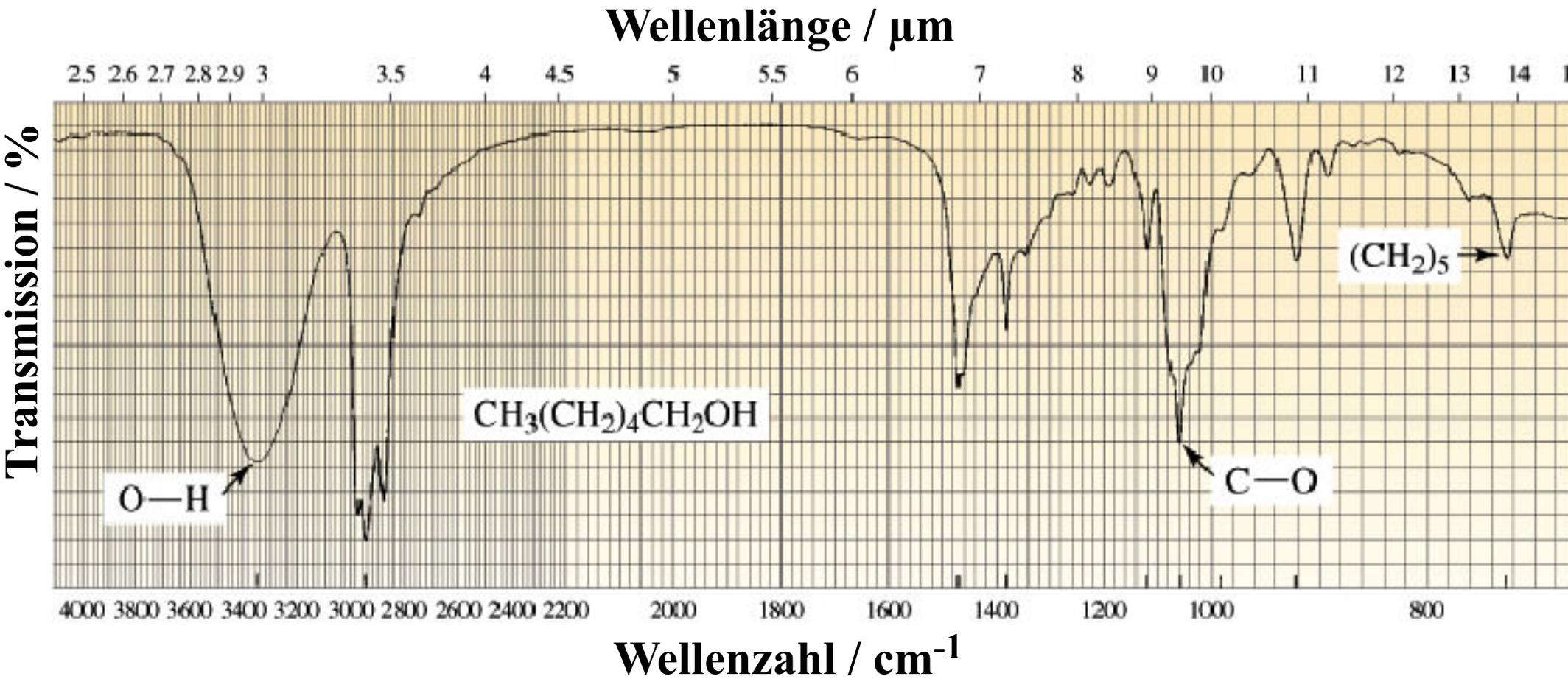
Zuordnung

-H Streck-  
schwingungen

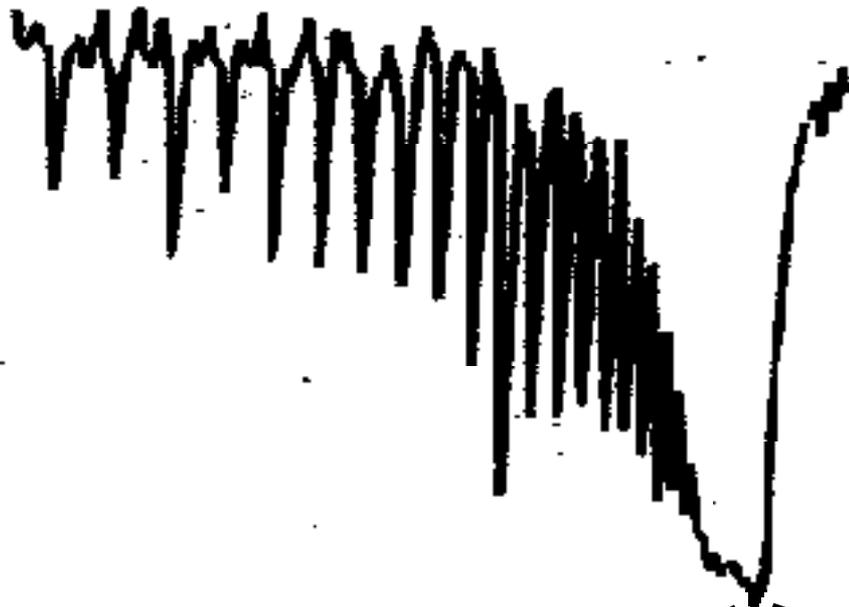


- O-H st
- N-H st
- C-O st
- C-H st
- C-H st
- S-H st
- B-H st
- X-Y st
- X-Y-Z st
- P-H st
- Si-H st
- C=O st
- C-N st
- C=O st
- N-H s
- B-O st
- C-N st
- C-F st
- Si-O st
- C-O st
- P-O st
- C-S st
- P-O st
- N-O st
- C-H s
- C-H s
- Si-O st

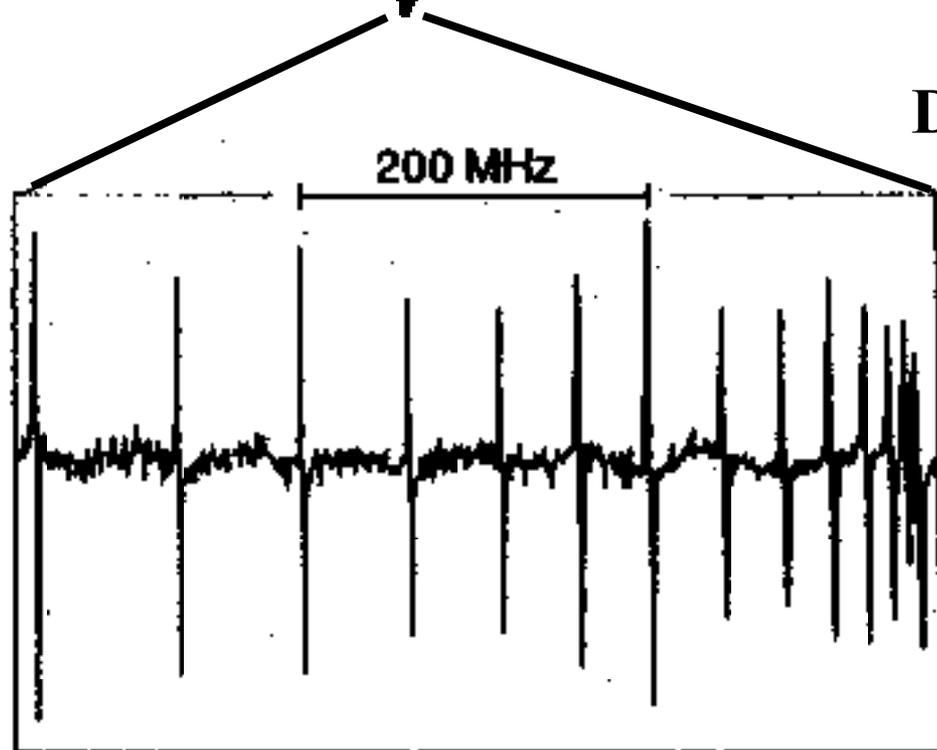
# IR-Spektrum



# Hochauflösende IR Spektroskopie



molekulare Absorptionsbande von O<sub>3</sub>  
FT-IR Spektrometer



Detail mit hoher Auflösung  
(FIR-Laser)

1.580 950

1.581 760 MHz

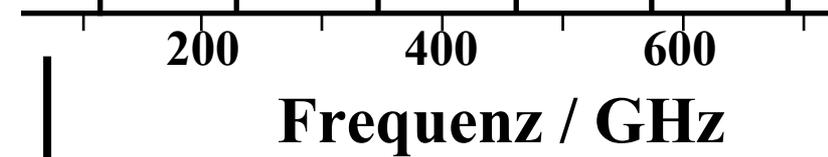
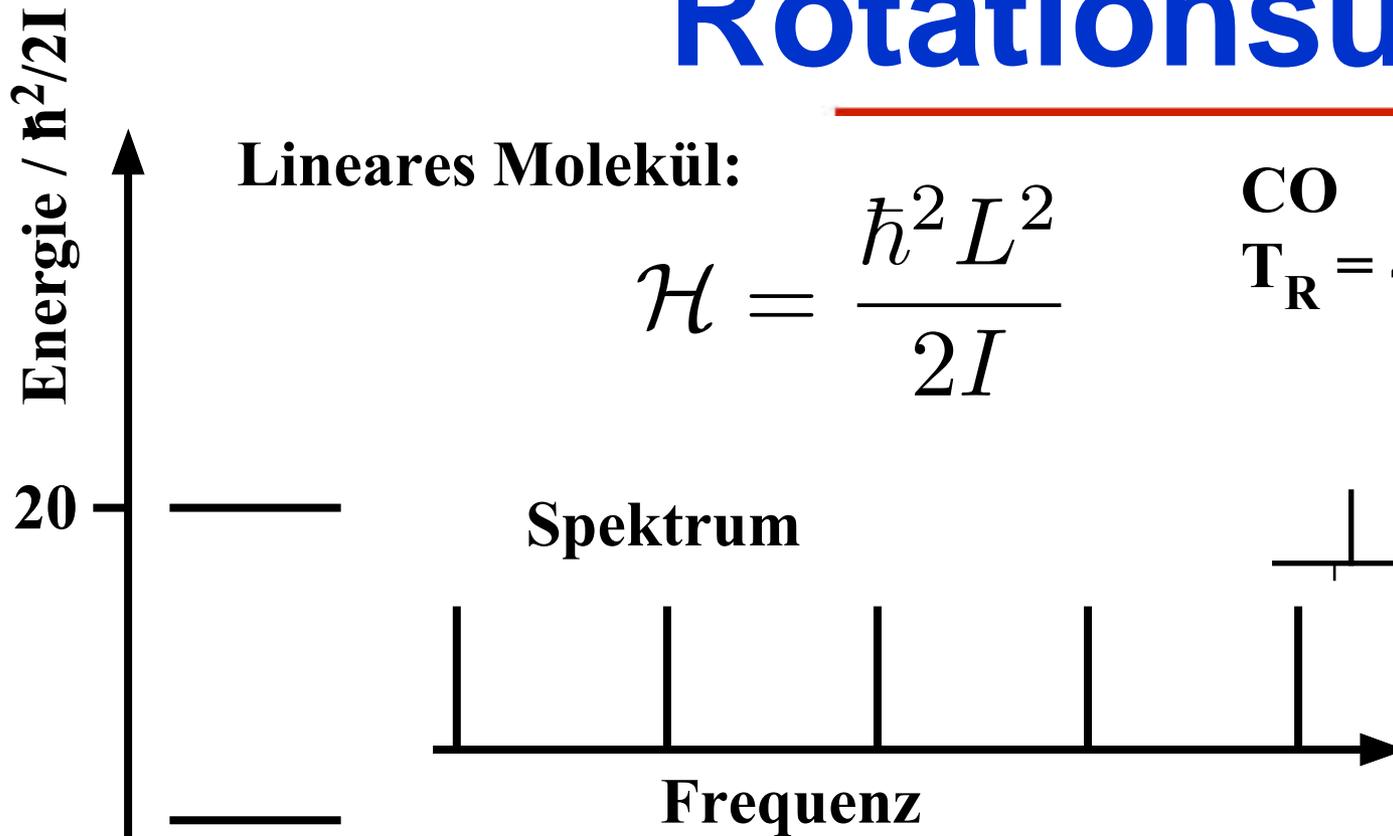
# Rotationsübergänge

Lineares Molekül:

$$\mathcal{H} = \frac{\hbar^2 L^2}{2I}$$

CO

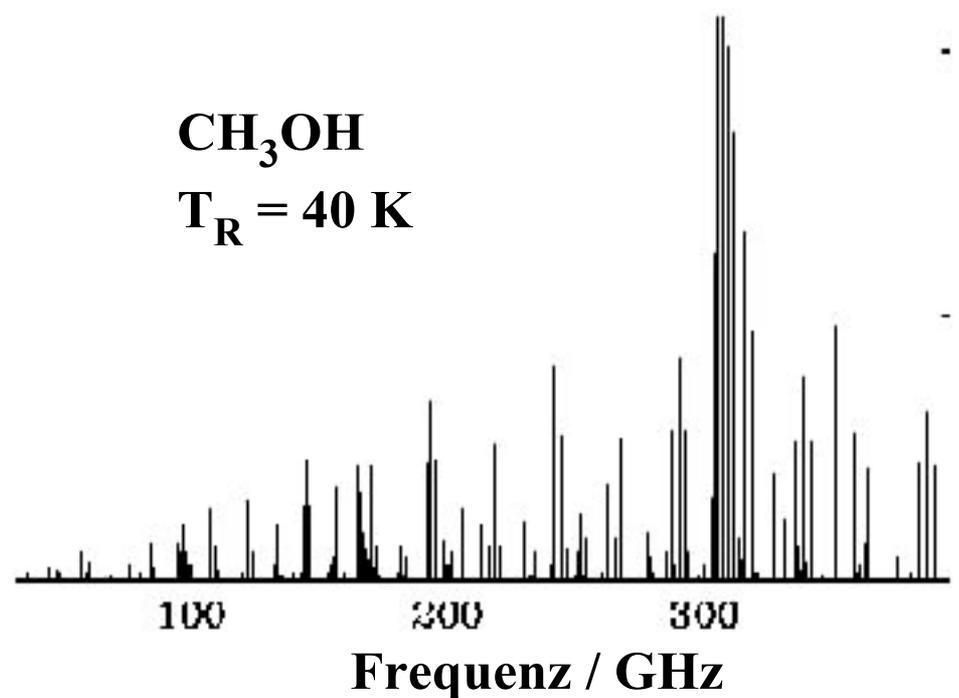
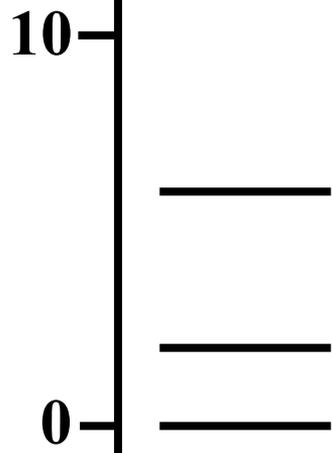
$T_R = 40\text{K}$



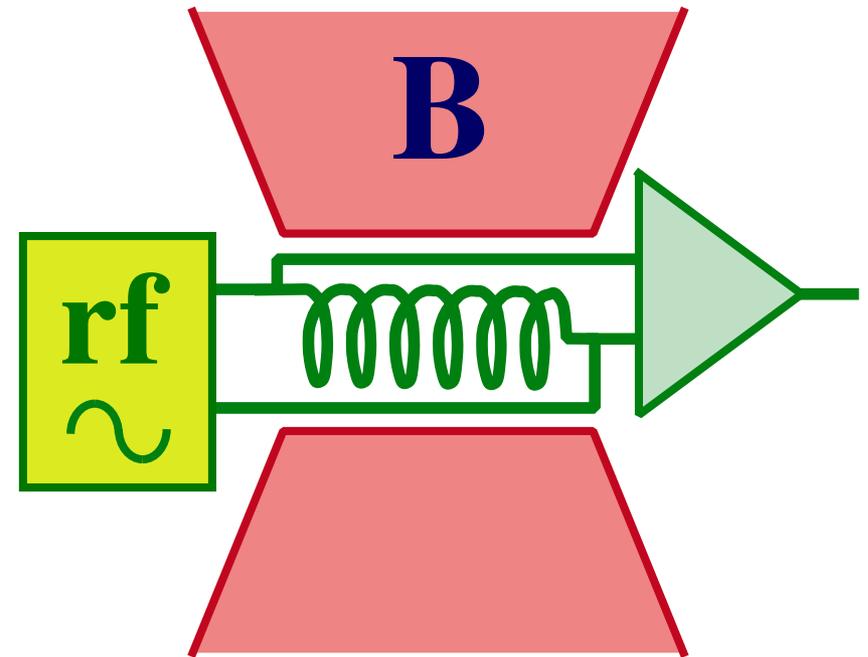
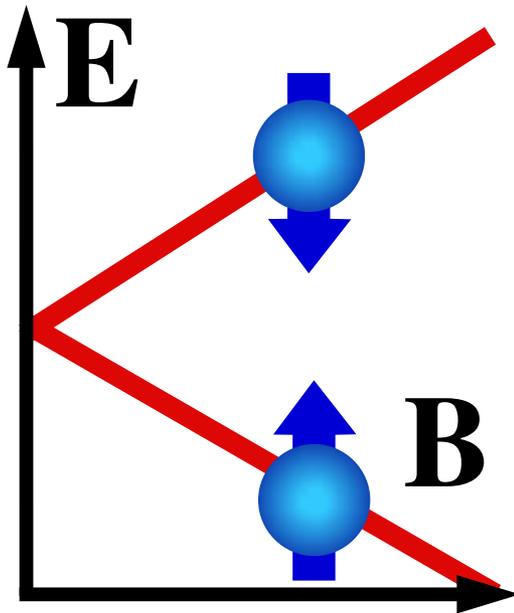
Asymmetrisches Molekül:

CH<sub>3</sub>OH

$T_R = 40\text{K}$



# Magnetische Resonanz



## Prinzip

Die magnetische Resonanz erzeugt und misst Übergänge zwischen unterschiedlichen Spinzuständen, die durch ein externes Magnetfeld aufgespalten werden.

## Ausführung

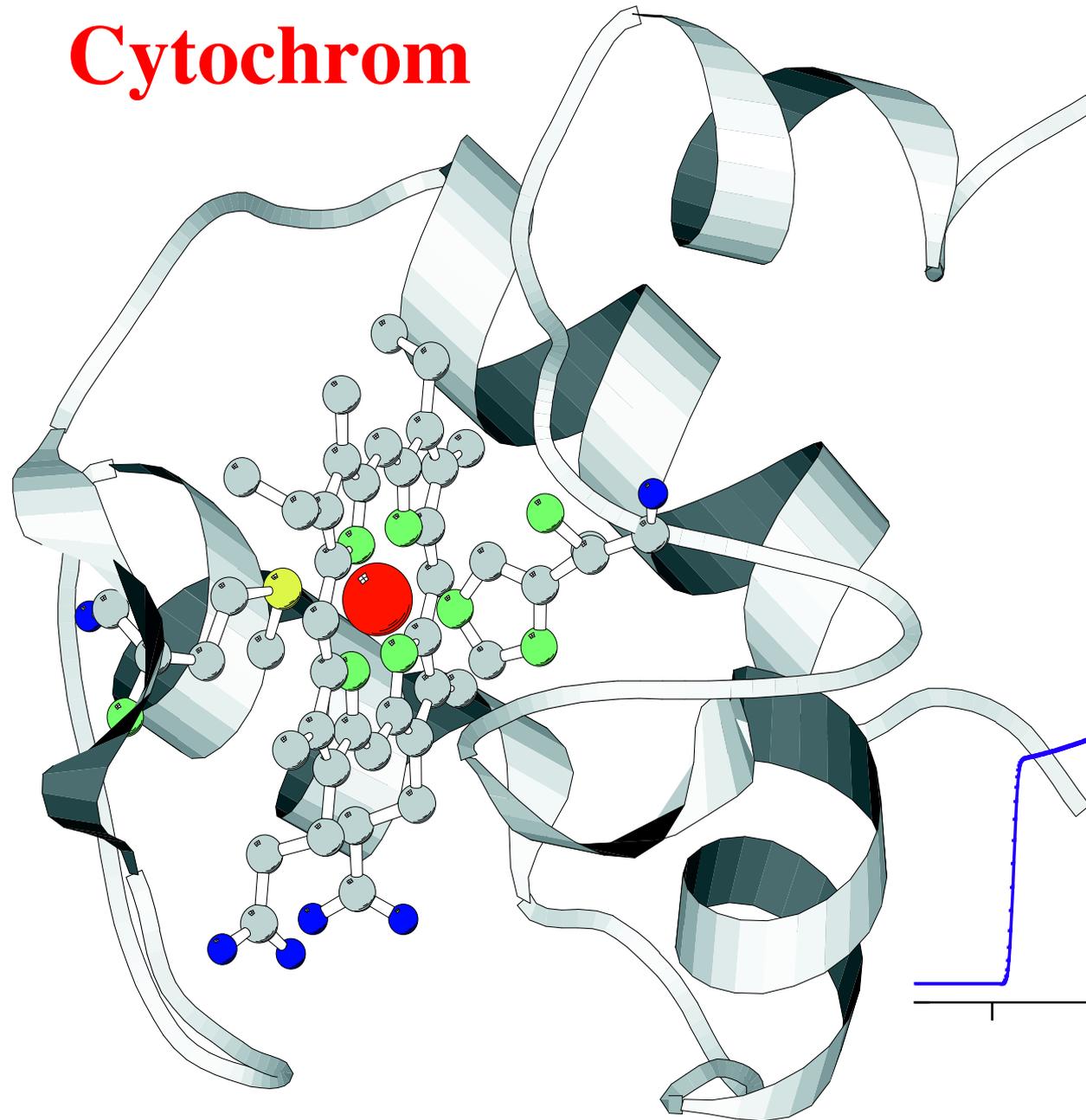
Statisches Magnetfeld

rf-Feld

~10 MHz - 100 GHz

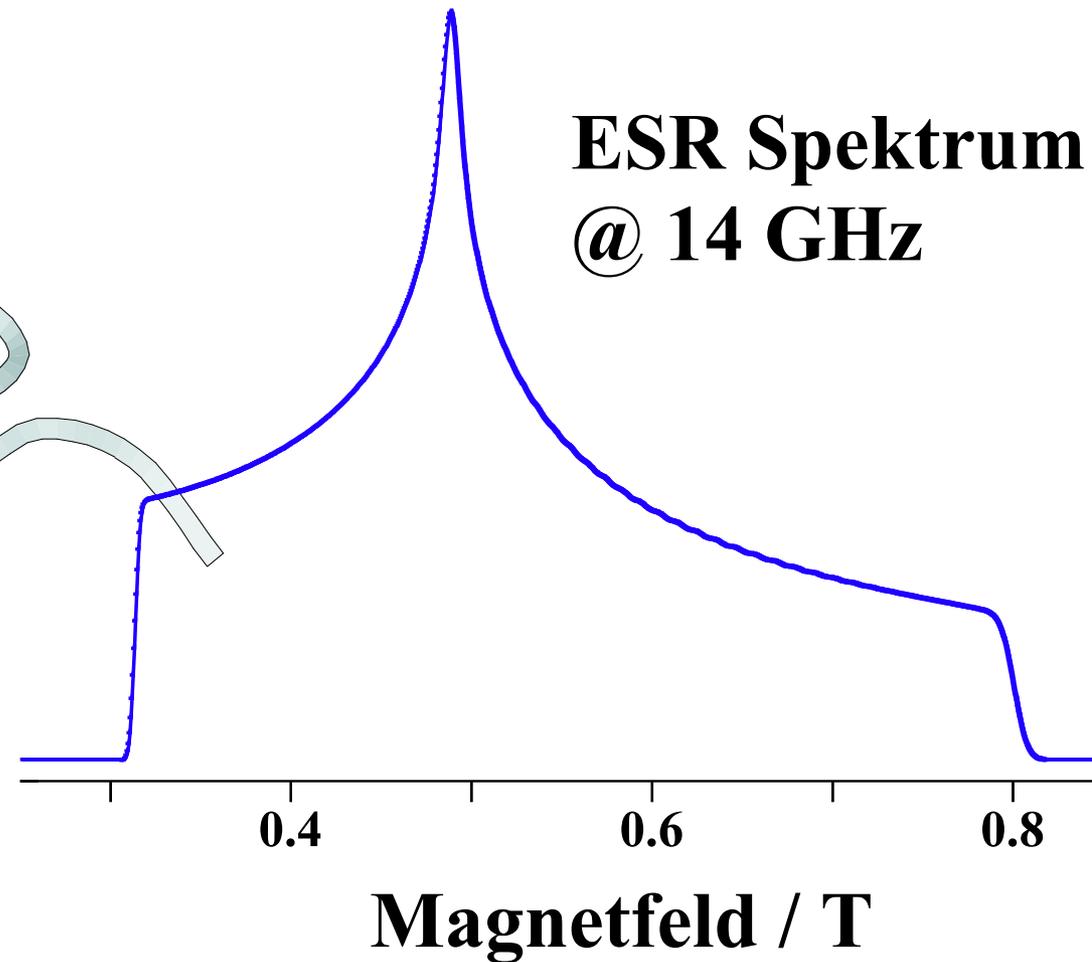
# Elektronenspinresonanz

## Cytochrom

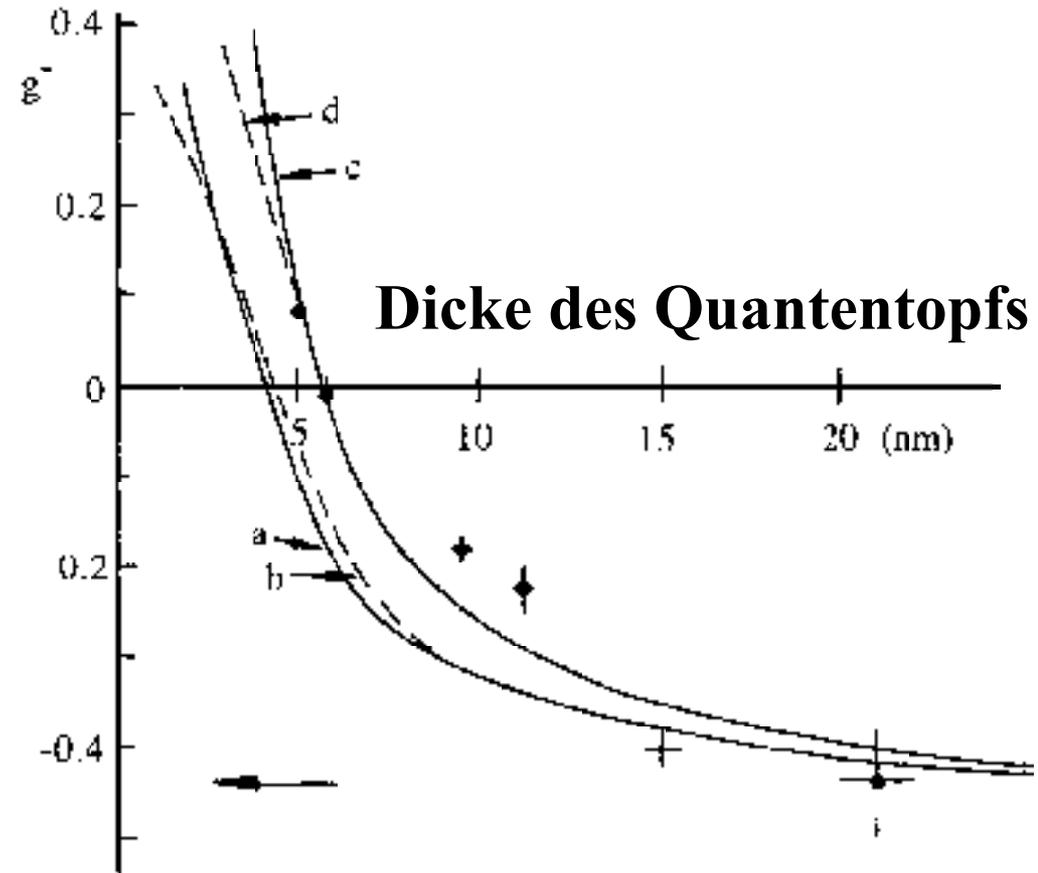
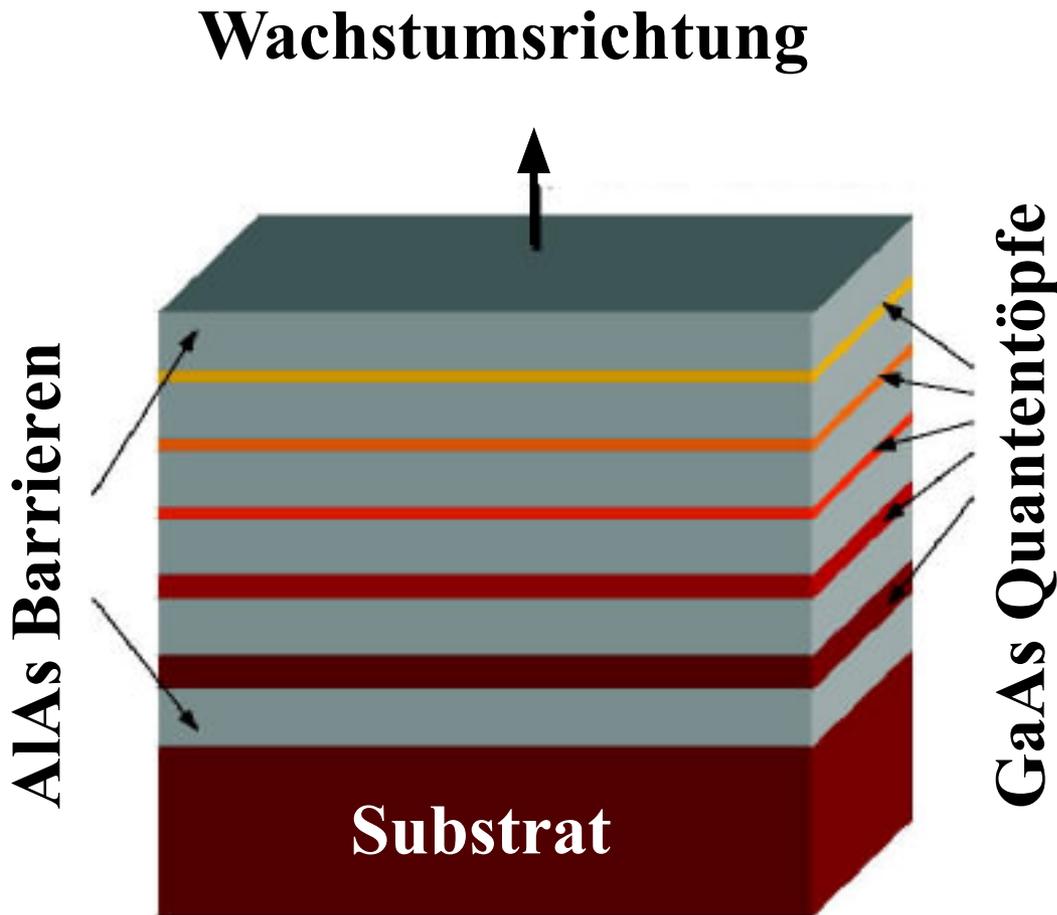


ungeordnetes System:  
Orientierungsmittel

ESR Spektrum  
@ 14 GHz



# GaAs Quantenfilme



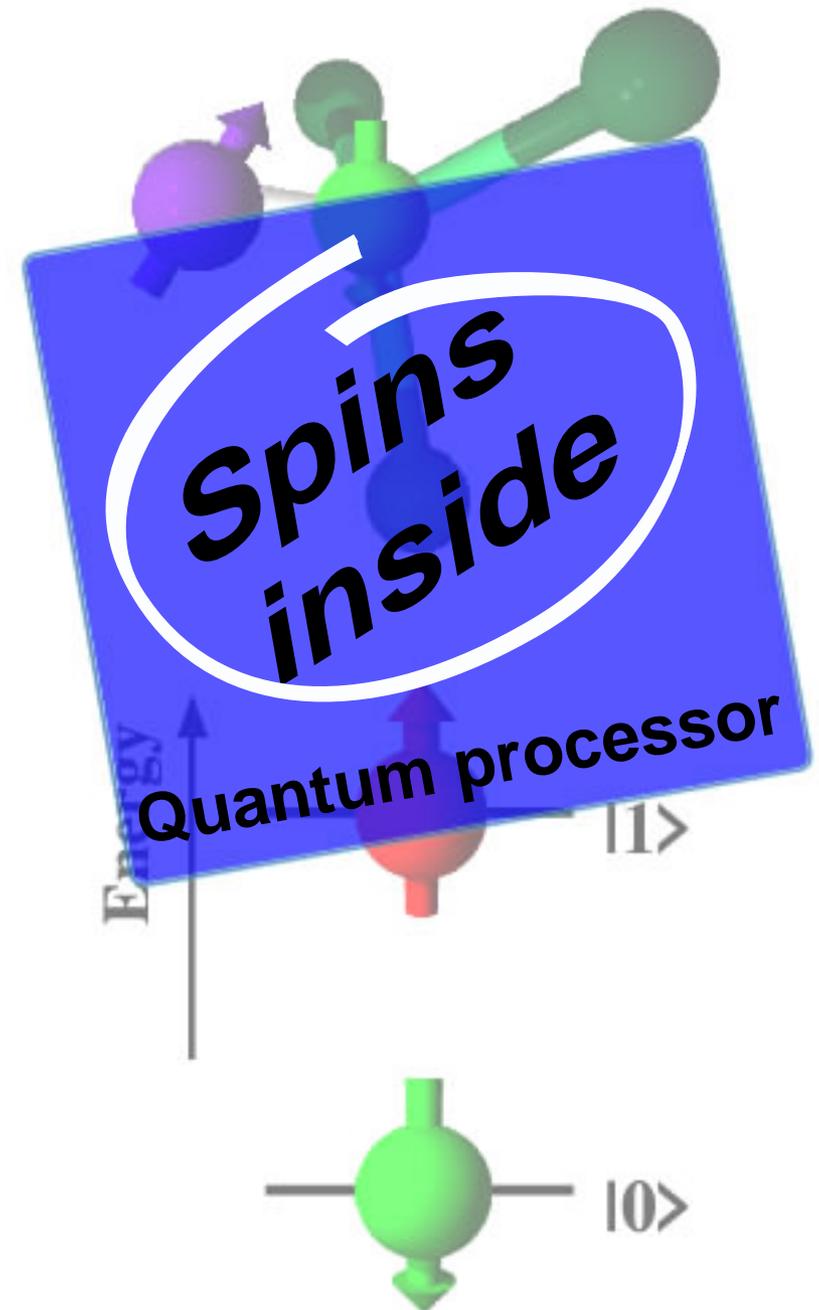
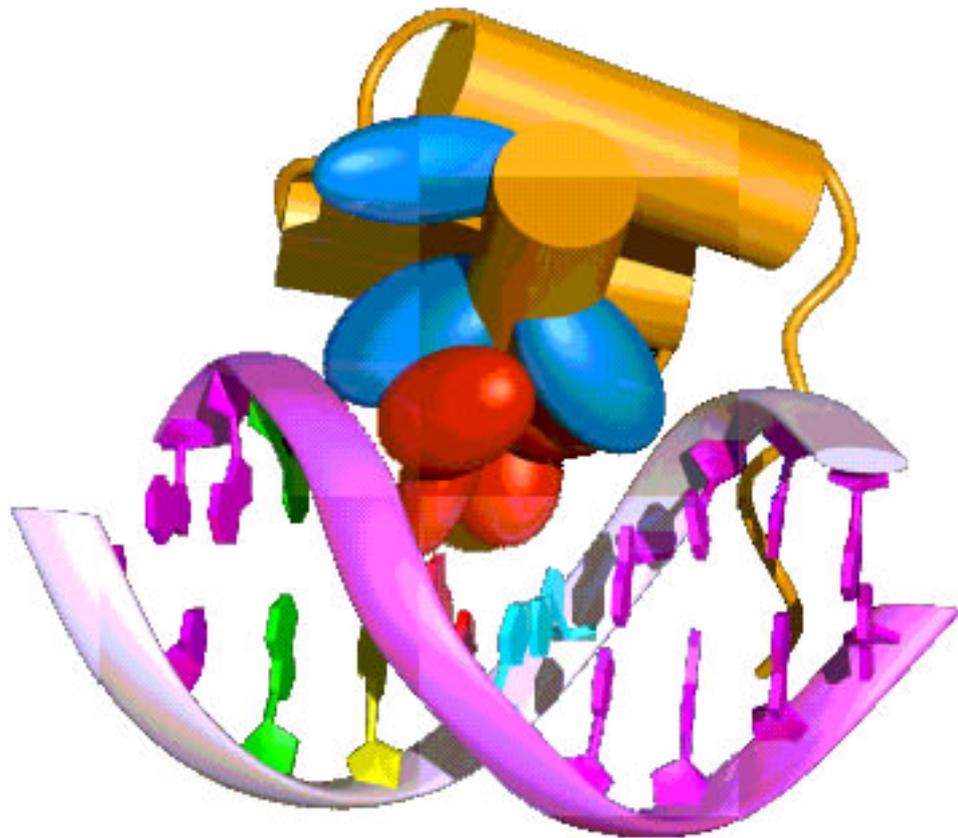
*Snelling et al., PRB 44, 11345 (1991).*

# Kernspinrsonanz (NMR)



# Anwendungen der NMR

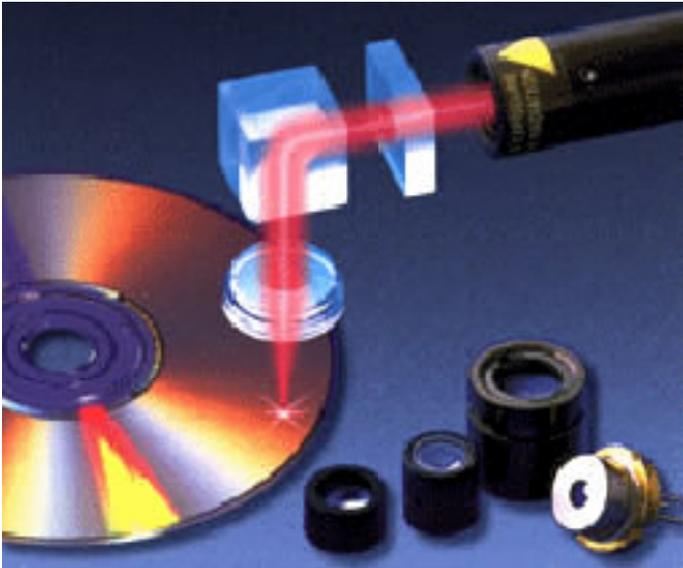
## Struktur und Dynamik von Molekülen



# Kernspintomographie (MRI)

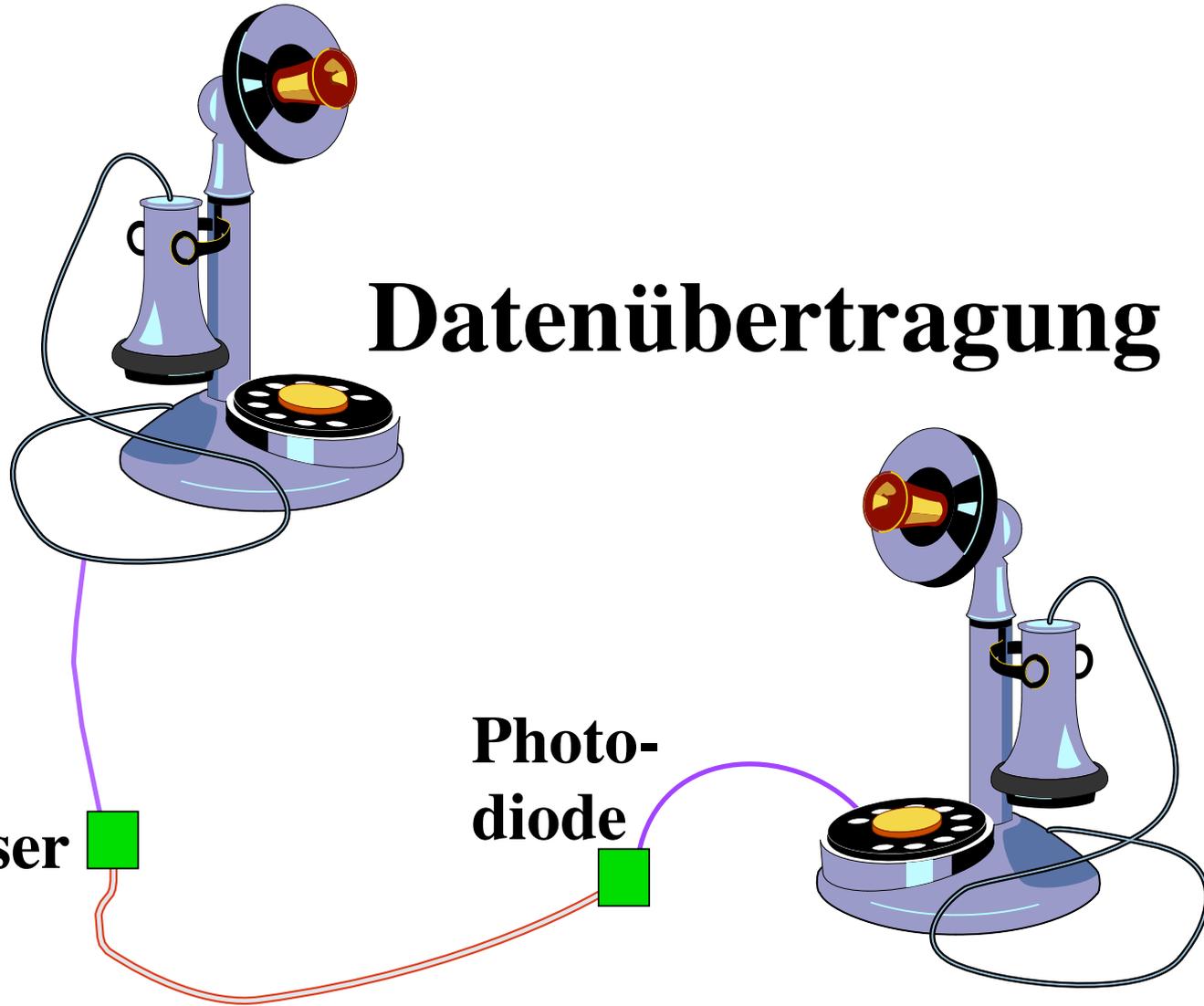


**CD, DVD**

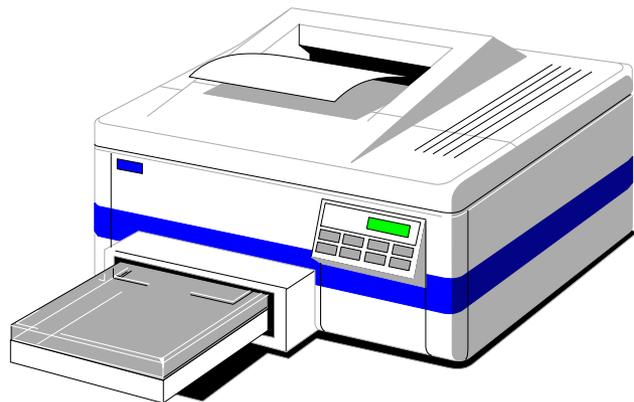


# **Laser Anwendungen**

**Datenübertragung**



**Laserdrucker**



**Laser**

**Photo-  
diode**

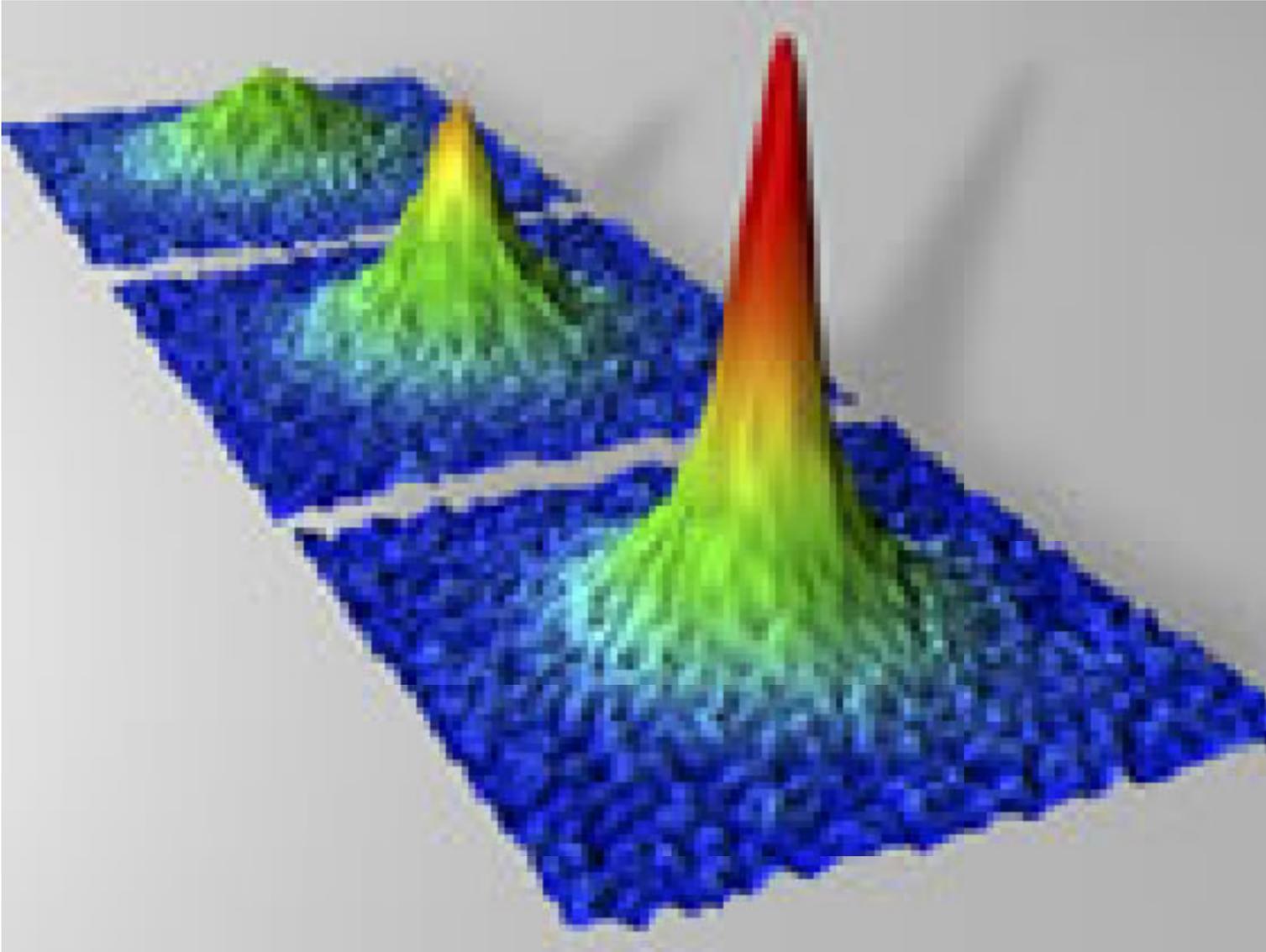
**Materialverarbeitung  
Unterhaltung**

**Medizin  
Messtechnik**

# Weitere Anwendungen von Lasern

---

## Kalte Atome, BEC



# Gravitationswellen

